

Key

## Electron Configuration Evaluation

In the space below, write the unabbreviated electron configurations of the following elements:

- 1) sodium  $1s^2 2s^2 2p^6 3s^1$
- 2) iron  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^6$
- 3) bromine  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^5$
- 4) barium  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2$
- 5) neptunium  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^6 7s^2 5f^4$

In the space below, write the abbreviated electron configurations of the following elements:

- 6) cobalt  $[Ar] 4s^2 3d^7$
- 7) silver  $[Kr] 5s^2 4d^9$
- 8) tellurium  $[Kr] 5s^2 4d^{10} 5p^4$
- 9) radium  $[Rn] 7s^2$
- 10) lawrencium  $[Rn] 7s^2 5f^{14}$

Determine what elements are denoted by the following electron configurations:

- 11)  $1s^2 2s^2 2p^6 3s^2 3p^4$  S
- 12)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$  Rb
- 13)  $[Kr] 5s^2 4d^{10} 5p^3$  As
- 14)  $[Xe] 6s^2 4f^{14} 5d^0$  Os
- 15)  $[Rn] 7s^2 6d^1 5f^{11}$  Fm

Determine which of the following electron configurations are not valid:

- 16)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 4d^{10} 4p^5$  Br
- 17)  $1s^2 2s^2 2p^6 3s^2 3d^1$  Mn
- 18)  $[Ra] 7s^2 5f^8$  Bk
- 19)  $[Kr] 5s^2 4d^{10} 5p^5$  I
- 20)  $[Xe]$  Xe

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**WJ Hussar**



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## **Table of Contents Chemistry Guided Practice Problems Answer Key**

1. Understanding the eBook Chemistry Guided Practice Problems Answer Key
  - The Rise of Digital Reading Chemistry Guided Practice Problems Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Guided Practice Problems Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Guided Practice Problems Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Guided Practice Problems Answer Key
  - Personalized Recommendations
  - Chemistry Guided Practice Problems Answer Key User Reviews and Ratings
  - Chemistry Guided Practice Problems Answer Key and Bestseller Lists
5. Accessing Chemistry Guided Practice Problems Answer Key Free and Paid eBooks
  - Chemistry Guided Practice Problems Answer Key Public Domain eBooks

- Chemistry Guided Practice Problems Answer Key eBook Subscription Services
- Chemistry Guided Practice Problems Answer Key Budget-Friendly Options
- 6. Navigating Chemistry Guided Practice Problems Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Guided Practice Problems Answer Key Compatibility with Devices
  - Chemistry Guided Practice Problems Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Guided Practice Problems Answer Key
  - Highlighting and Note-Taking Chemistry Guided Practice Problems Answer Key
  - Interactive Elements Chemistry Guided Practice Problems Answer Key
- 8. Staying Engaged with Chemistry Guided Practice Problems Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Guided Practice Problems Answer Key
- 9. Balancing eBooks and Physical Books Chemistry Guided Practice Problems Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Guided Practice Problems Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Guided Practice Problems Answer Key
  - Setting Reading Goals Chemistry Guided Practice Problems Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Guided Practice Problems Answer Key
  - Fact-Checking eBook Content of Chemistry Guided Practice Problems Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks

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#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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