

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY**Solutions****Section 15.1 What are solutions?**

In your textbook, read about the characteristic of solutions.

Underline the terms below just now to complete the passage.

solvent	liquid	solvent
insoluble	solid	solvent

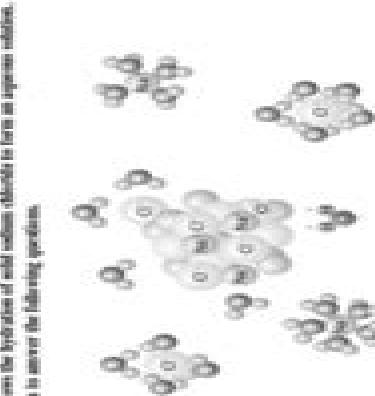
After reading (1) _____ solution _____ of solute is dissolved in solvent.

(2) _____ solute _____ is the (3) _____ solute _____ and solute is the

(4) _____ solution. A solution that does not dissolve is
 (5) _____ insoluble. (6) _____ liquid _____ solution is the most common
 type of solution. If one liquid is soluble in another liquid, such as water and oil, water
 can dissolve in (7) _____ insoluble _____ because if one liquid is insoluble in another
 the liquids are (8) _____ immiscible.

Read about solubility in aqueous solution in your textbook.

Underline the questions to answer the following questions.



1. Indicate in solute is when the solute is in a solution. What is solute?

Solute is the process of surrounding solute particles with solvent particles to form a solution.

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Density, Matter and Change • Chapter 11

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Section 15.2 Solubility**STUDY GUIDE FOR CONTENT MASTERY****Section 15.3 Factors that Affect Solubility****STUDY GUIDE FOR CONTENT MASTERY**

16. An sodium chloride dissolved in water. What happens to the sodium and chloride ions? The sodium and chloride ions are separated and surrounded by the water molecules.

17. Explain the attraction of the water molecules around the sodium ions and chloride ions. Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atom). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

18. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know? The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solute particles to scatter.

19. In a glass cup, add the rest of solute may be dissolved, stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent.

In your textbook, read about heat of solution, solubility and factors that affect solubility for each solute and solvents, with true or false.

True 16. The overall energy change that occurs when a solute dissolves is called the heat of solution.

False 17. Solubility is a property of the solute, amount of water the solute is in a given amount of solvent at a specified temperature and pressure.

False 18. Solute particles in a glass of the solution are in motion due to the heat of solution.

True 19. In a saturated solution, solute and solvents are in equilibrium.

True 20. Additional solute can be dissolved in an unsaturated solution.

True 21. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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privileged communities have re imagined and restructured blended education to enhance teaching and learning for underprivileged communities This book aims to address blended learning solutions across institutional program course and activity levels The chapters will cover a variety of learning environments from rural settings to less developed countries and more and explore the programs and courses designed to improve student success and accessibility in diverse student populations This book is ideally intended for teachers administrators teacher educators practitioners stakeholders researchers academicians and students who are interested in blended learning opportunities in less privileged settings and to underserved and marginalized populations *Clinical Nursing Skills and Techniques - E-Book* Anne G. Perry,Patricia A. Potter,Wendy R. Ostendorf,Nancy Laplante,2024-01-16 Learn the clinical nursing skills you will use every day and prepare for success on the Next Generation NCLEX Examination Clinical Nursing Skills Techniques 11th Edition provides clear step by step guidelines to more than 200 basic intermediate and advanced skills With more than 1 200 full color illustrations a nursing process framework and a focus on evidence based practice this manual helps you learn to think critically ask the right questions at the right time and make timely decisions Written by a respected team of experts this trusted text is the bestselling nursing skills book on the market Comprehensive coverage includes more than 200 basic intermediate and advanced nursing skills and procedures Rationales for each step within skills explain the why as well as the how of each skill and include citations from the current literature Clinical Judgments alert you to key steps that affect patient outcomes and help you modify care as needed to meet individual patient needs UNIQUE Unexpected Outcomes and Related Interventions sections highlight what might go wrong and how to appropriately intervene Clinical Review Questions at the end of each chapter provides case based review questions that focus on issues such as managing conflict care prioritization patient safety and decision making More than 1 200 full color photos and drawings help you visualize concepts and procedures Nursing process format provides a consistent presentation that helps you apply the process while learning each skill NEW All new Clinical Judgment in Nursing Practice chapter incorporates concepts of the NCSBN clinical judgment model Updated evidence based literature is incorporated throughout the skills NEW End of chapter questions and end of unit unfolding case studies provide optimal preparation for the Next Generation NCLEX NGN

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High Leverage Practices for Inclusive Classrooms James McLeskey,Lawrence Maheady,Bonnie Billingsley,Mary T. Brownell,Timothy J. Lewis,2022-03-30 High Leverage Practices for Inclusive Classrooms

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Engineering Education John Heywood,2006-01-24 A synthesis of nearly 2 000 articles to help make engineers better educators While a significant body of knowledge has evolved in the field of engineering education over the years much of the published information has been restricted to scholarly journals and has not found a broad audience This publication rectifies that situation by reviewing the findings of nearly 2 000 scholarly articles to help engineers become better educators devise more effective curricula and be more effective leaders and advocates in curriculum and research development The author s first objective is to provide an illustrative review of research and development in engineering education since 1960 His second objective is with the examples given to encourage the practice of classroom assessment and research and his third objective is to promote the idea of curriculum leadership The publication is divided into four main parts Part I demonstrates how the underpinnings of education history philosophy psychology sociology determine the aims and objectives of the curriculum and the curriculum s internal structure which integrates assessment content teaching and learning Part II focuses on the curriculum itself considering such key issues as content organization trends and change A chapter on interdisciplinary and integrated study and a chapter on project and problem based models of curriculum are included Part III examines problem solving creativity and design Part IV delves into teaching assessment and evaluation beginning with a chapter on the lecture cooperative learning and teamwork The book ends with a brief insightful forecast of the future of engineering education Because this is a practical tool and reference for engineers each chapter is self contained and may be read independently of the others Unlike other works in engineering education which are generally intended for educational researchers this publication is written not only for researchers in the field of engineering education but also for all engineers who teach All readers acquire a host of practical skills and knowledge in the fields of learning philosophy sociology and history as they specifically apply to the process of engineering curriculum improvement and evaluation

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