

How to Name Chemical Compounds

Is it ionic or covalent?
or does it contain a polyatomic ion followed by H⁺?

Ionic compound

[Cation's name] [Anion's name]

- metals**
- same as element name
ex. Al³⁺: aluminum
 - if transition metal⁺, include charge in Roman numeral
ex. Cu²⁺: copper (II)

- poly-anions**
- just use the name!
ex. NH₄⁺: ammonium
 - just use H₃O⁺
ex. SO₄²⁻: sulfate

* exceptions (no Roman numeral):

silver ion: silver Ag⁺

gold ion: gold(3+) Au³⁺

cadmium ion: cadmium Cd²⁺

Molecular compound

[H]-[1st element name] [H]-[2nd element root]-"ide"

Acids
root and
naming

↑ skip if "mono-"

↑ skip if 1st element is H

Stem/Prefixes

- 1: mono-
- 2: di-
- 3: tri-
- 4: tetra-
- 5: penta-
- 6: hexa-
- 7: hepta-
- 8: octa-
- 9: nona-
- 10: deca-



Acids

Drop all H⁺, and look at the name of the remaining anion.

Acid	Anion	Acid	Anion
HCl	chloride	H ₂ SO ₄	sulfate
H ₂ CO ₃	carbonate	H ₃ PO ₄	phosphate

Chemistry Study Guide How Elements Form Compound

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