

Section Review 3.1

Part A Completion

1. accuracy	6. absolute value
2. precision	7. 100%
3. experimental value	8. scientific notation
4. error	9. known
5. accepted value	10. estimated

Part B True-False

11. ST 13. AT
12. AT

Part C Matching

14. b	16. a	18. d
15. c	17. f	19. e

Part D Questions and Problems

20. a. 3
b. 4
21. a. 4.35×10^2 mL
b. 8.4 cm^2

c. 2
e. 2.1×10^2

Section Review 3.2

Part A Completion

1. metric	6. gram
2. seven	7. Celsius
3. meter	8. kelvin
4. liter	9. joule/calorie
5. weight	10. calorie/joule

Part B True-False

11. NT 13. AT
12. NT 14. NT

Part C Matching

15. k	20. l	25. j
16. e	21. d	26. f
17. m	22. c	27. b
18. g	23. h	
19. n	24. i	

Part D Questions and Problems

28. Volume = $1.8 \text{ cm} \times 8.8 \text{ cm} \times 30.5 \text{ cm}$
= $4.8 \times 10^3 \text{ cm}^3$

29. $^{\circ}\text{C} = \text{K} - 273 = 20 - 273 = -253^{\circ}\text{C}$

30. a. $\text{kg} \cdot 10^3$
b. $\text{kg} \cdot 10^3$

c. $\text{m} \cdot 10^{-3}$
d. $\text{m} \cdot 10^{-3}$

Section Review 3.3

Part A Completion

1. one (unity)	6. steps
2. conversion factor	7. conversion factor
3. remains the same	8. denominator
4. dimensional	9. units
analysis	10. unknown
5. known	

Part B True-False

11. NT 13. AT 15. AT
12. ST 14. NT

Part C Questions and Problems

16. a. $125 \text{ g} \times \frac{1 \text{ kg}}{1000 \text{ g}} = 0.125 \text{ kg}$
= $1.25 \times 10^{-1} \text{ kg}$

b. $0.12 \text{ L} \times \frac{1000 \text{ mL}}{1 \text{ L}} = 1.2 \times 10^2 \text{ mL}$

17. $\frac{1.0 \text{ } \mu\text{m}}{1500 \text{ cells}} \times \frac{2.54 \text{ } \mu\text{m}}{1 \text{ } \mu\text{m}} \times \frac{1 \text{ } \mu\text{m}}{100 \text{ } \mu\text{m}} \times \frac{10^6}{1 \text{ } \mu\text{m}}$
= $17 \text{ } \mu\text{m}/\text{cell}$

18. $186\,000 \text{ spades} \times \frac{1 \text{ s}}{1 \text{ spade}} \times \frac{1.61 \text{ km}}{10^6 \text{ spades}} \times \frac{1 \text{ gaffe}}{1 \text{ gaffe}}$
= $2.99 \times 10^{-1} \text{ km}$

19. $\frac{1 \text{ } \mu\text{m}}{1 \text{ } \mu\text{m}}$

Section Review 3.4

Part A Completion

1. density	3. composition
2. intensive	

Part B True-False

4. ST 5. ST

Part C Questions and Problems

6. $32.1 \text{ mL} \times \frac{1 \text{ cm}^3}{1 \text{ mL}} = 32.1 \text{ cm}^3$

7. Density = $\frac{\text{mass}}{\text{volume}} = \frac{127 \text{ g}}{32.1 \text{ cm}^3} = 3.96 \text{ g/cm}^3$

7. $1.00 \text{ L} \times \frac{1000 \text{ cm}^3}{1 \text{ L}} = 1000 \text{ cm}^3$

$1.58 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} = 1580 \text{ g}$

Density = $\frac{\text{mass}}{\text{volume}} = \frac{1580 \text{ g}}{1000 \text{ cm}^3} = 1.58 \text{ g/cm}^3$

Database Test Study Guide Answers

Charles J. Brooks

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