

Section Review 3.1

Part A Completion

- | | |
|-----------------------|------------------------|
| 1. accuracy | 6. absolute value |
| 2. precision | 7. 100% |
| 3. experimental value | 8. scientific notation |
| 4. error | 9. known |
| 5. accepted value | 10. estimated |

Part B True-False

- | | |
|--------|--------|
| 11. ST | 13. AT |
| 12. AT | |

Part C Matching

- | | | |
|-------|-------|-------|
| 14. b | 16. a | 18. d |
| 15. e | 17. f | 19. c |

Part D Questions and Problems

- | | |
|------------------------------|----------------------|
| 20. a. 3 | c. 2 |
| b. 4 | |
| 21. a. 4.35×10^3 mL | e. 2.1×10^2 |
| b. 8.4 cm^3 | |

Section Review 3.2

Part A Completion

- | | |
|-----------|-------------------|
| 1. metric | 6. gram |
| 2. seven | 7. Celsius |
| 3. meter | 8. kelvin |
| 4. liter | 9. joule/calorie |
| 5. weight | 10. calorie/joule |

Part B True-False

- | | |
|--------|--------|
| 11. NT | 13. AT |
| 12. NT | 14. NT |

Part C Matching

- | | | |
|-------|-------|-------|
| 15. k | 20. i | 25. j |
| 16. e | 21. d | 26. f |
| 17. m | 22. c | 27. b |
| 18. g | 23. h | |
| 19. a | 24. i | |

Part D Questions and Problems

28. Volume = $1.8 \text{ cm} \times 8.8 \text{ cm} \times 30.5 \text{ cm}$
 $= 4.8 \times 10^2 \text{ cm}^3$
29. $^{\circ}\text{C} = \text{K} - 273 = 20 - 273 = -253^{\circ}\text{C}$
- | | |
|---------------|--------------|
| 30. a. 10^2 | c. 10^{-3} |
| b. 10^3 | d. 10^{-2} |

Section Review 3.3

Part A Completion

- | | |
|-------------------------|----------------------|
| 1. one (unity) | 6. steps |
| 2. conversion factor | 7. conversion factor |
| 3. remains the same | 8. denominator |
| 4. dimensional analysis | 9. units |
| 5. known | 10. unknown |

Part B True-False

- | | | |
|--------|--------|--------|
| 11. NT | 13. AT | 15. AT |
| 12. ST | 14. NT | |

Part C Questions and Problems

16. a. $125 \text{ g} \times \frac{1 \text{ kg}}{1000 \text{ g}} = 0.125 \text{ kg}$
 $= 1.25 \times 10^{-1} \text{ kg}$
- b. $0.12 \text{ L} \times \frac{1000 \text{ mL}}{1 \text{ L}} = 1.2 \times 10^2 \text{ mL}$
17. $\frac{1.0 \text{ } \mu\text{m}}{1500 \text{ cells}} \times \frac{2.54 \text{ cm}}{1 \text{ } \mu\text{m}} \times \frac{1 \text{ m}}{100 \text{ cm}} \times \frac{10^6}{1 \text{ } \mu\text{m}}$
 $= 17 \text{ } \mu\text{m/cell}$
18. $\frac{185 \text{ 000 mph}}{1 \text{ s}} \times \frac{1 \text{ s}}{10^6 \text{ } \mu\text{s}} \times \frac{1.61 \text{ km}}{1 \text{ mile}}$
 $= \frac{2.99 \times 10^{-1} \text{ km}}{1 \text{ } \mu\text{s}}$

Section Review 3.4

Part A Completion

- | | |
|--------------|----------------|
| 1. density | 3. composition |
| 2. intensive | |

Part B True-False

- | | |
|-------|-------|
| 4. ST | 5. ST |
|-------|-------|

Part C Questions and Problems

6. $32.1 \text{ mL} \times \frac{1 \text{ cm}^3}{1 \text{ mL}} = 3.21 \text{ cm}^3$
- Density = $\frac{\text{mass}}{\text{volume}} = \frac{127 \text{ g}}{32.1 \text{ cm}^3} = 3.96 \text{ g/cm}^3$
7. $1.00 \text{ L} \times \frac{1000 \text{ cm}^3}{1 \text{ L}} = 1000 \text{ cm}^3$
- $1.58 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} = 1580 \text{ g}$
- Density = $\frac{\text{mass}}{\text{volume}} = \frac{1580 \text{ g}}{1000 \text{ cm}^3} = 1.58 \text{ g/cm}^3$

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Charles J. Brooks



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