

Lewis Dot Structure	Chemical Formula	Name
$\text{:C}\equiv\text{O:}$	CO	carbon monoxide
$\text{:}\ddot{\text{O}}=\text{C}=\ddot{\text{O}}\text{:}$	CO ₂	carbon dioxide
	BF ₃	boron trifluoride
$\text{:N}\equiv\text{N}-\ddot{\text{O}}\text{:}$	N ₂ O	dinitrogen <u>monoxide</u>
	N ₂ O ₃	dinitrogen trioxide

CO₂



2 is a **subscript number**. It appears slightly below the regular line and stands for the number of atoms of the element that comes before it in a compound. If there is not subscript, it means there is only one element.

Covalent Compounds Study Guide

Jill Meryl Levy



Covalent Compounds Study Guide:

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