

CHAPTER 7 REVIEW***Chemical Formulas and Chemical Compounds*****SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. c In a Stock name such as iron(III) sulfate, the roman numeral tells us _____.
 (a) how many atoms of Fe are in one formula unit
 (b) how many sulfate ions can be attached to the iron atom
 (c) the charge on each Fe ion
 (d) the total positive charge of the formula unit

2. c The result of changing a subscript in a correctly written chemical formula is to _____.
 (a) change the number of moles represented by the formula
 (b) change the charges on the other ions in the compound
 (c) change the formula so that it no longer represents the compound it previously represented
 (d) have no effect on the formula

3. The explosive TNT has the molecular formula $\text{C}_7\text{H}_5(\text{NO}_2)_3$.
4 elements a. How many elements make up this compound?
6 oxygen atoms b. How many oxygen atoms are present in one molecule of $\text{C}_7\text{H}_5(\text{NO}_2)_3$?
21 atoms c. How many atoms in total are present in one molecule of $\text{C}_7\text{H}_5(\text{NO}_2)_3$?
 4.2×10^{24} atoms d. How many atoms are present in a sample of 2×10^{23} molecules of $\text{C}_7\text{H}_5(\text{NO}_2)_3$?

4. How many atoms are present in each of these formula units?
11 atoms a. $\text{Ca}(\text{HCO}_3)_2$
45 atoms b. $\text{C}_{12}\text{H}_{22}\text{O}_{11}$
10 atoms c. $\text{Fe}(\text{ClO}_3)_3$
9 atoms d. $\text{Fe}(\text{ClO}_3)_2$

5. N_2O_5 a. What is the formula for the compound dinitrogen pentoxide?
carbon(IV) sulfide b. What is the Stock name for the covalent compound CS_2 ?
 H_2SO_4 c. What is the formula for sulfuric acid?
phosphoric acid d. What is the name for the acid H_3PO_4 ?

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Table of Contents Chemical Names And Formulas Study Guide Answers

1. Understanding the eBook Chemical Names And Formulas Study Guide Answers
 - The Rise of Digital Reading Chemical Names And Formulas Study Guide Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Names And Formulas Study Guide Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Names And Formulas Study Guide Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Names And Formulas Study Guide Answers
 - Personalized Recommendations
 - Chemical Names And Formulas Study Guide Answers User Reviews and Ratings
 - Chemical Names And Formulas Study Guide Answers and Bestseller Lists
5. Accessing Chemical Names And Formulas Study Guide Answers Free and Paid eBooks
 - Chemical Names And Formulas Study Guide Answers Public Domain eBooks
 - Chemical Names And Formulas Study Guide Answers eBook Subscription Services

- Chemical Names And Formulas Study Guide Answers Budget-Friendly Options
- 6. Navigating Chemical Names And Formulas Study Guide Answers eBook Formats
 - ePUB, PDF, MOBI, and More
 - Chemical Names And Formulas Study Guide Answers Compatibility with Devices
 - Chemical Names And Formulas Study Guide Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Names And Formulas Study Guide Answers
 - Highlighting and Note-Taking Chemical Names And Formulas Study Guide Answers
 - Interactive Elements Chemical Names And Formulas Study Guide Answers
- 8. Staying Engaged with Chemical Names And Formulas Study Guide Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Names And Formulas Study Guide Answers
- 9. Balancing eBooks and Physical Books Chemical Names And Formulas Study Guide Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Names And Formulas Study Guide Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemical Names And Formulas Study Guide Answers
 - Setting Reading Goals Chemical Names And Formulas Study Guide Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemical Names And Formulas Study Guide Answers
 - Fact-Checking eBook Content of Chemical Names And Formulas Study Guide Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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