

# Challenging Calculations - Study-Guide

4. How many molecules are there in 2.48 moles of  $\text{CO}_2$ ?

It is 2.48 x  $10^{24}$  molecules

5. How many formula units are there in 2.5 moles of  $\text{CaBr}_2$ ?

It is 2.5 x  $10^{24}$  fu

6. How many molecules are there in 8 moles of isopropyl alcohol,  $\text{C}_3\text{H}_8\text{O}$ ?

It is molecules

7. How many moles are there in 2

It is 20.4 g/mol

8. How many moles are there in 2

It is 2.5 g/mol

9. What is the mass formula units of  $\text{CaCl}_2$ ?

It is 110.9 g/mol

10. What is the number of moles in 2.5 grams of  $\text{NaCl}$ ?

It is 0.043 moles of  $\text{NaCl}$

11. How many formula units are there in 2.5 g. What is the number of moles in 2.5 g. of  $\text{NaCl}$ ?

It is 2.5 g/mol

It is 2.5 g/mol

12. If 2.5 g of sodium chloride is dissolved in 2.5 g of water, what is the mass of the solution?

It is 2.5 g/mol

13. What is the mass formula units of  $\text{CaCl}_2$ ?

It is 110.9 g/mol

14. What is the number of moles in 2.5 g of  $\text{NaCl}$ ?

It is 0.043 moles

# Chemical Quantities Study Guide Answers

**Julie R. Frentrup**



## Chemical Quantities Study Guide Answers:

**Study Guide/Selected Solutions Manual** Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book

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