

Explain:

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1. Write down the following reactions in the form of word equations:
- Reaction 1:  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- Reaction 2:  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
- Reaction 3:  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
- Reaction 4:  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$

2. Write down the following reactions in the form of word equations:
- Reaction 5:  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
- Reaction 6:  $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$
- Reaction 7:  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
- Reaction 8:  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$

3. Write down the following reactions in the form of word equations:
- Reaction 9:  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
- Reaction 10:  $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$
- Reaction 11:  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
- Reaction 12:  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$

4. Write down the following reactions in the form of word equations:
- Reaction 13:  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
- Reaction 14:  $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$
- Reaction 15:  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
- Reaction 16:  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$

5. Write down the following reactions in the form of word equations:
- Reaction 17:  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
- Reaction 18:  $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$
- Reaction 19:  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
- Reaction 20:  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$

6. Write down the following reactions in the form of word equations:
- Reaction 21:  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
- Reaction 22:  $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$
- Reaction 23:  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
- Reaction 24:  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$

7. Write down the following reactions in the form of word equations:
- Reaction 25:  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
- Reaction 26:  $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$
- Reaction 27:  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
- Reaction 28:  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$

8. Write down the following reactions in the form of word equations:
- Reaction 29:  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
- Reaction 30:  $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$
- Reaction 31:  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
- Reaction 32:  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$

9. Write down the following reactions in the form of word equations:
- Reaction 33:  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
- Reaction 34:  $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$
- Reaction 35:  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
- Reaction 36:  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$

10. Write down the following reactions in the form of word equations:
- Reaction 37:  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
- Reaction 38:  $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$
- Reaction 39:  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
- Reaction 40:  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$

11. Write down the following reactions in the form of word equations:
- Reaction 41:  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
- Reaction 42:  $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$
- Reaction 43:  $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
- Reaction 44:  $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$

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**Katherine J. Denniston, Denniston  
Katherine**



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