

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write yes if evidence of a chemical reaction is present. Write no if there is no evidence of a chemical reaction.

yes 1. A tomato smells rotten.

no 2. A drinking glass breaks into smaller pieces.

no 3. A piece of ice melts.

yes 4. Drain cleaner is mixed with water and the solution becomes warm.

yes 5. Candle wax burns.

yes/no 6. Molten candle wax solidifies.

yes 7. Green leaves turn yellow and red as the seasons change.

yes 8. Baking powder produces a gas that makes a cake rise.

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(g)	(g)
reactant	product	(g)	(g)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is also (9) reactant, oxygen is also (9) reactant, and water vapor is also (9) product. In a chemical equation for this reaction, also (12) arrow is used to separate hydrogen and oxygen from water vapor and energy. Also (13) plus sign is used to separate the symbols for hydrogen and oxygen. Also (14) (g) symbol is used to tell the state of hydrogen in the reaction, also (15) (g) symbol is used for the state of oxygen, and also (16) (g) symbol is used for the state of water vapor.

Chemistry Chemical Reactions Study Guide Key

Milin Kurup

Chemistry Chemical Reactions Study Guide Key:

Study Guide to Organic Chemical Reactions Milin Kurup, 2020-01-06 This Organic Study Guide and workbook was created by the author in hopes to help students have an organized collection of all reactions studied in Organic 1 and Organic 2. With multiple reactions in different categories an organized study guide can help students focus on certain reactions as needed. Many students in this class also have hopes of using this study guide as an MCAT review guide this organized set can be useful for students to review before class exams and standardized test. With comprehensive notes reactions and tips this study guide will help everyone succeed. This was created by a renowned student Milin Kurup at the University of Florida double majoring in Microbiology and Cognitive and Behavioral Neuroscience. As a student of Professor Dr Laura Peterson UF Chemistry Organic Division this guide included all notes a series of 140 chemical reactions and key mechanisms necessary for a comprehensive understanding of the subject.

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Morris Hein,Susan Arena,2003-09-03

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