

## Achievement

**Chapter Test B****Chapter: Chemical Equations and Reactions**

Directions: In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

1. The generation of a relatively unstable, short-lived compound by an alpha-disintegration reaction is called the **transmutation** of an atom.

- a. fusion
- b. fission
- c. radioactive interaction
- d. beta decay

2. The products of a chemical equation, in the presence of the catalyst, are:

- a. ions
- b. molecules
- c. atoms and/or molecules
- d. numbers of particles

3. In a chemical equation, the greatest (biggest) molecule that can be cut into smaller pieces is the **reactant**.

- a. result
- b. arrangement for reaction
- c. reactant
- d. reactants

4. The tendency for a radioactive nucleus to decay into a more stable nucleus is called **radioactive decay**.

- a. nuclear fusion
- b. nuclear fission
- c. nuclear fission
- d. nuclear fission and fusion

5. The advantage of a radioactive isotope for medical diagnosis is that:

- a. it is emitted from a radioactive source
- b. it has a short half-life
- c. it is a stable isotope
- d. it remains radioactive for many millions of years

6. The advantage for a radioactive isotope:

- a. it is emitted from a radioactive source
- b. it has a short half-life
- c. it is a stable isotope
- d. it remains radioactive for many millions of years

7. If one of  $^{238}\text{U}$  in Radium-226 would, in the time available, decay:

- a.  $^{238}\text{U}$  would become  $^{234}\text{Ra}$  in radioactive
- b.  $^{238}\text{U}$  would become  $^{234}\text{Ra}$  in radioactive
- c.  $^{238}\text{U}$  would become radioactive
- d.  $^{238}\text{U}$  would remain radioactive

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