

CHEMISTRY TEST

NAME: _____

INTRODUCTORY CHEMISTRY CONCEPTS

DATE: _____

Directions: For each of the following questions, choose the number that best answers the question and place it on your answer sheet.**Directions:** For each of the following questions, all major steps must be shown to receive full credit.**Introductory Chemistry Concepts****Level 1****Multiple Choice**

1. A chemical change always results characteristically in a change of
1) state; 2) form; 3) identity; 4) size; 5) shape.

ANS: 3

2. Which of the following is a chemical change?
1) evaporation of water 2) distillation of water 3) freezing of water
4) electrolysis of water

ANS: 4

3. Chemical changes always involve 1) the absorption of energy; 2) the liberation
of energy; 3) either the absorption or liberation of energy; 4) changes of state.
ANS: 3

4. One physical property of sugar is its 1) acidity; 2) reactivity;
3) flammability; 4) plasma qualities; 5) solubility.

ANS: 5

5. The majority of chemical changes are 1) autothermic; 2) endothermic;
3) isothermal; 4) neithermic; 5) exothermic.

ANS: 5

6. The statement which describes a physical property of the element oxygen is
1) oxygen supports the burning of paper; 2) oxygen is needed for the rusting of
iron; 3) oxygen must be present for food to decay; 4) oxygen has a density
of 1.43 grams per liter; 5) oxygen is a component of water.

ANS: 4

7. In any chemical reaction energy is 1) always absorbed; 2) always
released; 3) released as heat; 4) usually not involved; 5) either
absorbed or released.

ANS: 5

8. Which of the following is not a chemical change?
1) the dissolving of zinc by HCl 2) the exploding of gasoline
3) the corroding of a copper roof 4) the boiling of water

ANS: 4

9. One chemical property of a substance is a description of its 1) odor; 2) density;
3) combustibility; 4) solubility; 5) none of these.

ANS: 3

10. A new substance is formed as a result of a 1) physical change; 2) chemical

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Answer Key

R R Bowker Publishing, Bowker

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Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change **Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH₄ from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement [A Level Chemistry MCQ \(Multiple Choice Questions\)](#) Arshad Iqbal, 2019-06-18 The A Level Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF A Level Chemistry MCQ PDF Download Quiz Questions Chapter 1 28 Practice Tests with Answer Key IGCSE GCE Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs A Level Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests A Level Chemistry MCQ PDF book helps to practice test questions from exam prep notes The A Level Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs A Level Chemistry Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved quiz questions and answers on chapters Alcohols and esters atomic structure and theory benzene chemical compound carbonyl compounds carboxylic acids acyl compounds chemical bonding chemistry of life electrode potential electrons in atoms enthalpy change equilibrium group IV groups II and VII halogenoalkanes hydrocarbons introduction to organic chemistry ionic equilibria lattice energy moles and equations nitrogen and sulfur organic and nitrogen compounds periodicity polymerization rates of reaction reaction kinetics redox reactions and electrolysis states of matter transition elements tests for college and university revision guide A Level Chemistry Quiz Questions and Answers PDF free download eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The book IGCSE GCE Chemistry MCQs Chapter 1 28 PDF includes high school question papers to review practice tests for exams A Level Chemistry Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for IGCSE NEET MCAT GRE GMAT SAT ACT competitive exam A Level Chemistry Mock Tests Chapter 1 28 eBook covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Alcohols and Esters MCQ Chapter 2 Atomic Structure and Theory MCQ Chapter 3 Benzene Chemical Compound MCQ Chapter 4 Carbonyl Compounds MCQ Chapter 5 Carboxylic Acids and Acyl Compounds MCQ Chapter 6 Chemical Bonding MCQ Chapter 7 Chemistry of Life MCQ Chapter 8 Electrode Potential MCQ Chapter 9 Electrons

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