

Arrhenius Definitions: When reacting with water

- Acids release hydrogen ions (ex: HNO₃)
- Bases release hydroxide ions (ex: NaOH)
- Salts are ionic compounds that release neither
hydrogen ions nor hydroxide ions

Practice: Write "A" for acid, "B" for base, and "S" for salt.

A	B	S
B	A	

Solution	Acid (A), Base (B), or Salt (S)	Reacts with Mg (R or NR)	Conductivity (C or NC)
HNO ₃	A	R	C
NaOH	B	NR	C
NaCl	S	NR	NC
H ₂ SO ₄	A	R	C
HCl	A	R	C
NH ₄ OH	B	NR	C
KNO ₃	S	NR	NC
KOH	B	NR	C
HC ₂ H ₃ O ₂	A	R	C

Chemistry Note Taking Guide Episode 1101 Answers

Wolfgang Guggemos

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