

**CHEMFILE MINI-GUIDE TO PROBLEM SOLVING****General Plan for Solving Dilution Problems**

$$1 \quad M_1 V_1 = M_2 V_2$$

Rearrange the equation  
 $M_1 V_1 = M_2 V_2$   
 algebraically to solve for  
 the unknown quantity.

**2**

The equation used to  
 calculate the unknown quantity  
 will be one of the following four:

$$V_2 = \frac{M_1 V_1}{M_2}, \quad M_2 = \frac{M_1 V_1}{V_2}, \quad V_1 = \frac{M_2 V_2}{M_1}, \quad M_1 = \frac{M_2 V_2}{V_1}$$

Substitute each of the  
 known quantities for its  
 symbol, and calculate.

**3**

Unknown  
 molarity or volume

**SAMPLE PROBLEM 1**

What is the molarity of a solution that is made by diluting 50.00 mL of a 4.74 M solution of HCl to 250.00 mL?

**SOLUTION****1. ANALYZE**

• *What is given in the problem?*

the molarity of the stock solution,  
 the volume used to dilute, and the  
 volume of the diluted solution

• *What are you asked to find?*

the molarity of the diluted solution

**Items****Data**

Concentration of the stock solution ( $M_1$ )

4.74 M HCl

Volume of stock solution used ( $V_1$ )

50.00 mL

Volume of diluted solution ( $V_2$ )

250.00 mL

Concentration of the diluted solution ( $M_2$ )

?

# Chemfile Mini Guide To Problem Solving

**Julia Schneider**

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