

CHEMISTRY CHEAT SHEET

EQUILIBRIUM

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Chemical equilibrium is a state of balance in which the forward and reverse reactions occur at the same rate. The equilibrium constant, K , is the ratio of the equilibrium concentrations of products to reactants.

General Equilibrium:
A reversible reaction in which reactants are converted into products, and products are converted back into reactants.

Reversible Equilibrium: A reaction in which the forward and reverse reactions occur simultaneously at the same rate.

Chemical Equilibrium: A state of equilibrium in which the concentrations of all species in a reaction mixture remain constant over time.

Equilibrium Constant: A numerical expression of equilibrium that shows the relative amounts of products and reactants at equilibrium.

Equilibrium Constant, K : A constant that represents the ratio of the equilibrium concentrations of products to reactants.

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Chemistry Equilibrium Study Guide

Leo J. Malone

Chemistry Equilibrium Study Guide:

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Chemistry, Study Guide Bernice G. Segal, Peter S. Shenkin, 1989-02-14

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Study Guide for General Chemistry and College Chemistry, Eighth Editions by Holtzclaw and Robinson Norman E. Griswold, Henry Fuller Holtzclaw, 1988 **Basic Concepts of Chemistry, Study Guide** Leo J.

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