

DENSITY WORKBOOK**Question:** What is Density?

Answer: **Density** is a measure of how much mass is contained in a given unit volume. Put simply, if mass is a measure of how much 'stuff' there is in an object; **density** is a measure of how tightly that 'stuff' is packed together. The principle of density was discovered by the Greek scientist Archimedes.

*Overview of Density Worksheet***Key formulas/Concepts:**

- Density** - Density = Mass divided by Volume ($D = M/V$). The mass of an object is 25 grams. The volume of an object is 5 cm³. $D = 25g/5cm^3 = 5 g/cm^3$.

$$\text{density} = \frac{\text{mass}}{\text{volume}}$$

- Volume of a rectangular shaped object/cube** - Length X Width X Height ($L \times W \times H$).

Example - A cube has a length of 3 cm. (Note - a cube has six equal sides). $V = 3 \times 3 \times 3 = 27 \text{ cm}^3$. A piece of wood has a length of 10 cm, width of 18 cm, and height of 3 cm. $V = 10 \times 18 \times 3 = 540 \text{ cm}^3$.

- Volume of an irregularly shaped object** - Use a graduated cylinder filled with a fluid. Record the beginning amount of fluid (in mL - milliliters). Drop the object into the graduated cylinder. Record the level of the fluid with the object (in mL). To determine the volume: **Level of fluid with object (minus) beginning amount of fluid**. Example - a graduated cylinder is filled to the 30mL mark with water. You drop in a rock. The water level rises to 48mL. $V = 48\text{mL} - 30\text{mL} = 18\text{cm}^3$.

KEY NOTE

- ✓ A submerged object displaces a volume of liquid equal to the volume of the object.
- ✓ One milliliter (1 mL) of water has a volume of 1 cubic centimeter (1cm³).

Problems (Always Show Your Work and units - g, mL, cm³, g/cm³, g/mL):

_____ 1. $M = 34g$, $V = 17\text{cm}^3$

_____ 2. $M = 42g$, $V = 28\text{cm}^3$

_____ 3. You have a box that has a volume of 210 cm³ and mass of 42g.
What is its density?

Density Workbook Answer Guide

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