

## Calculating pH

How is pH related to the concentration of hydronium ions?

### Why?

In biology and other science courses pH is introduced as a way to quantify the acidity or basicity of a solution. This property can be measured using a pH probe or with an indicator paper strip that changes color at different pH values. But, what is actually being measured? We know that a pH of 7 is neutral, below 7 is acid, and above 7 is base, but why? What in the solution is the paper strip or probe actually reacting with?

### Model 1 – Ion Concentrations for Acids and Bases

Beaker	Solution	Acidic, Basic or Neutral?	$[H_3O^+]$	$[OH^-]$	$[H_3O^+] \times [OH^-]$
1	0.10 M HCl(aq)	Acidic	$1.0 \times 10^{-1}$ M	$1.0 \times 10^{-13}$ M	$1 \times 10^{-14}$
2	0.0010 M HCl(aq)	Acidic	$1.0 \times 10^{-3}$ M	$1.0 \times 10^{-11}$ M	$1 \times 10^{-14}$
3	0.000010 M HCl(aq)	Acidic	$1.0 \times 10^{-5}$ M	$1.0 \times 10^{-9}$ M	$1 \times 10^{-14}$
4	0.0000010 M HCl(aq)	Acidic	$1.0 \times 10^{-6}$ M	$1.0 \times 10^{-8}$ M	$1 \times 10^{-14}$
5	0.00000010 M HCl(aq)	Neutral	$1.0 \times 10^{-7}$ M	$1.0 \times 10^{-7}$ M	$1 \times 10^{-14}$
6	0.00000010 M NaOH(aq)	Neutral	$1.0 \times 10^{-7}$ M	$1.0 \times 10^{-7}$ M	$1 \times 10^{-14}$
7	0.0000010 M NaOH(aq)	Basic	$1.0 \times 10^{-8}$ M	$1.0 \times 10^{-6}$ M	$1 \times 10^{-14}$
8	0.00010 M NaOH(aq)	Basic	$1.0 \times 10^{-10}$ M	$1.0 \times 10^{-4}$ M	$1 \times 10^{-14}$
9	0.010 M NaOH(aq)	Basic	$1.0 \times 10^{-12}$ M	$1.0 \times 10^{-2}$ M	$1 \times 10^{-14}$
10	0.10 M NaOH(aq)	Basic	$1.0 \times 10^{-13}$ M	$1.0 \times 10^{-1}$ M	$1 \times 10^{-14}$

1. What does the symbol  $[H_3O^+]$  in Model 1 indicate?

*The molar concentration of hydronium ion.*

2. In Beaker 2, which ion has a higher concentration, hydronium ion or hydroxide ion?

*Hydronium ion.*

3. Describe how the concentration of hydronium ion was calculated for Beaker 3 in Model 1 from the concentration of the acid.

*Since HCl is a strong acid, all of the molecules ionize in water. Therefore, the  $H_3O^+$  concentration is equal to the initial acid concentration in the solution.*

4. Describe how the concentration of hydroxide ion was calculated for Beaker 8 in Model 1 from the concentration of the base.

*Since NaOH is a strong base, it completely dissociates in water. Therefore, the  $OH^-$  concentration is equal to the initial base concentration in the solution.*

# Calculating Ph Pogil Answer Key

**G. D. Peckham**



## Calculating Ph Pogil Answer Key:

**A Generalised Equation for Calculating PH** G. D. Peckham, 1984      **Ph** Hans Skrifvars, 1999      Measurement of PH Richard E. Lawn, Florence Elizabeth Prichard, 2003 Whether pH is being used to test a sample against a legal requirement or specification as part of an analytical method for monitoring and controlling a reaction as a process control in the chemical industry or for the environmental monitoring of waste and effluents it is important that all pH measurements are carried out in a logical and consistent manner paying careful attention to experimental procedures in order to obtain reliable results This guide provides scientists with the knowledge of how to do just that first by outlining the principles of pH measurement and buffer solutions pH meters and electrodes are then discussed including selection criteria and the care of electrodes Finally sections on making pH measurements and uncertainty are followed by a set of practical exercises Measurement of pH is one of the Practical Laboratory Skills Training Guides a series that aims to make achieving best practice easy These invaluable manuals will enable both experienced and inexperienced staff to get the essential basics of any experiment right simply by following the clear and easy to use instructions provided The guides are written by experienced scientists and include minimal theory plenty of practical exercises in order to assess competence and trouble shooting information Other titles are Measurement of Mass Measurement of Volume High Performance Liquid Chromatography and Gas Chromatography *Determination of PH* Roger Gordon Bates, 1973      Method of Determining PH by the Alkaline Absorption of Carbon Dioxide, 1992 A method for measuring the concentration of hydroxides in alkaline solutions in a remote location using the tendency of hydroxides to absorb carbon dioxide The method includes the passing of carbon dioxide over the surface of an alkaline solution in a remote tank before and after measurements of the carbon dioxide solution A comparison of the measurements yields the absorption fraction from which the hydroxide concentration can be calculated using a correlation of hydroxide or pH to absorption fraction

Calculating Ph Pogil Answer Key: Bestsellers in 2023 The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous engrossing novels enthralling the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the fascinating narratives that have charmed audiences this year. The Must-Read : Colleen Hoover "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Calculating Ph Pogil Answer Key : Delia Owens "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of compelling stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and suspenseful novel that will keep you wondering until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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