

Using Gas Law Equation:

At a temperature of 27°C and a pressure of 50.65 kPa, 60.0 mL of a gas is collected. What would the volume of the gas be at STP?

- STP is 273 K and a pressure of 101.3 kPa
- Set up the equation:

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

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