

Properties of Water



Water has the unique property of being a polar molecule. This is because the oxygen atom has a higher electron density than the hydrogen atoms, and the hydrogen atoms have a lower electron density than the oxygen atom.

Water has a high density compared to most other substances.



Water has a high surface tension compared to most other substances.



Water has a high vapor pressure compared to most other substances.

Water has a high boiling point compared to most other substances.

Water has a high melting point compared to most other substances.

Water has a high density compared to most other substances.

2.2 - WATER

2.2.1 - Chemical Characteristics of Water

Water is a polar molecule, which means that it has a positive charge on one side and a negative charge on the other side. This is because the oxygen atom has a higher electron density than the hydrogen atoms, and the hydrogen atoms have a lower electron density than the oxygen atom.



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Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh,2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application

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