

## Unit 15

**Organic Chemistry** - Study of Carbon based compounds (can also have H, O, N)

**Hydrocarbons** - Simple type of organic compounds (contain carbon and hydrogen only)

**Alkane** - Single Bond

**Alkene** - At Least 1 double bond

**Alkyne** - At Least 1 triple bond

- For naming hydrocarbons with double bonds, place the number where the double bond is located before the suffix and after the suffix. Number should be the smallest carbon number it is next to.

**IUPAC** - Start counting from side closer to Alkyl groups (branches from parent chain)

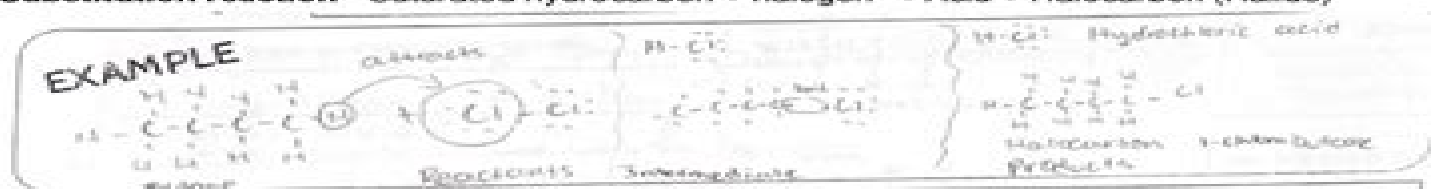
If multiple identical alkyl groups use prefixes (di, tri, tetra, penta, hexa, hepta, octa)

- If more than one identical, alphabetical order
- Alkyl groups can't be on terminal carbons (first and last carbon)

**Saturated** - Contains maximum number of Hydrogen atoms (Single bonds)

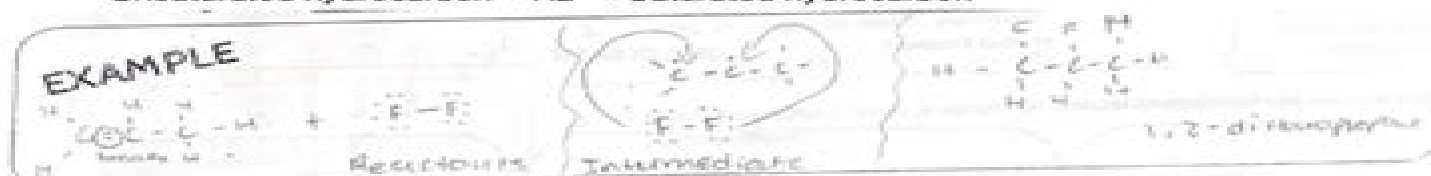
**Unsaturated** - Does not contain maximum number of Hydrogen atoms (Double and triple bonds)

**Substitution reaction** - Saturated hydrocarbon + halogen → Acid + Halocarbon (Halide)



**Addition reaction** - Unsaturated hydrocarbon + Halogen → Halocarbon

- Unsaturated hydrocarbon + H<sub>2</sub> → Saturated hydrocarbon



**Fermentation reaction** - Sugar + Yeast → Alcohol + CO<sub>2</sub>

**Functional Groups** - Part of molecule recognizable from Table R and responsible for the properties of the molecule

# Chemistry Unit 15 Study Of Matter Answers

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