

Ionic Bonding

The goal is to give students a picture of the entire and not the change! Subscripted right column, circled the subscripted column. Li^+ , O^{2-} , Mg^{2+} . This should clarify their goal of the changes to make.

Students understand ions and ionic bonding. Students must understand what happens in the process of the reaction. Li^+ is called the "lithium ion" and Mg^{2+} is called the "magnesium ion". Subscripted column must be correct a subscripted. The subscripted, just change the ending of the number "1" to "2". Subscript O^{2-} is correct subscript O^{2-} is subscripted.

Critical Thinking Questions

1. Write the correct subscripted structure and name for each of the following:
 Li^+ O^{2-} Mg^{2+} Al^{3+} S^{2-} Ca^{2+} K^+ Br^-

Correct: Li^+ is Li^+ , O^{2-} is O^{2-} , Mg^{2+} is Mg^{2+} , Al^{3+} is Al^{3+} , S^{2-} is S^{2-} , Ca^{2+} is Ca^{2+} , K^+ is K^+ , Br^- is Br^- .

Write structure, name, structure, and name.

Students must be able to write "lithium ion" and "oxygen ion" and "magnesium ion" and "aluminum ion" and "sulfur ion" and "calcium ion" and "potassium ion" and "bromine ion".

- How many electrons does each Li^+ have? How many electrons does O^{2-} have? How many electrons does Mg^{2+} have?
- How many electrons does each Al^{3+} have? How many electrons does S^{2-} have? How many electrons does Ca^{2+} have? How many electrons does K^+ have? How many electrons does Br^- have?
- How many electrons does each Li^+ have? How many electrons does O^{2-} have? How many electrons does Mg^{2+} have?

The subscripted column of the reaction is the subscripted column. Students are called "lithium ion", "oxygen ion", "magnesium ion", "aluminum ion", "sulfur ion", "calcium ion", "potassium ion", "bromine ion".

Critical Thinking Questions

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 Li^+ O^{2-} Mg^{2+} Al^{3+} S^{2-} Ca^{2+} K^+ Br^-
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 Li^+ O^{2-} Mg^{2+} Al^{3+} S^{2-} Ca^{2+} K^+ Br^-

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Sabine Zange

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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