

# Chemistry Study Guide

Periodic Table of the Elements																		
H	Li	Be																
Na	Mg																	
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Cs	Ba	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
Fr	Ra	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Fl	Uup	Lv	Uus	Uuo		
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
		Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		

## Objectives:

To understand the structure of an atom, and the nature + properties of subatomic particles.



## Matter:

Anything that has mass and occupies space.

Examples: air, gases, liquids, solidz!

## Atoms:

Atoms are the smallest, indivisible particle.

- Outer part → shell/orbit/energy level → carries the electronz (e-)
- Inner part → nucleus → houses the protonz (p+) and neutronz (n)
- Outer & Inner part → together these are the subatomic particles.

## Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p <sup>+</sup>	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e <sup>-</sup>	-0	-1	Orbit/Shell

# Chemistry Study Guide Section 2 Answers

**Gerard J. Criner, Gilbert E. D'Alonzo**

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Klein, Laurie S. Starkey, 2025-03-04 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Study Guide and Solutions Manual 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry

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**to accompany Basic Concepts of Chemistry, 9th Edition** Leo J. Malone, 2011-12-07 This is the Study Guide and Solutions Manual to accompany Malone's Basic Concepts of Chemistry

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