

> Chemistry Workbook answers

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Chapter 1

Exercise 1.1

Focus

- 1 A solid has a fixed *mass* and *volume* / *shape*.
A liquid has a fixed *mass* but its *shape* changes to match that of the container in which it is placed. A gas has no fixed *shape* or *volume*.
A gas completely fills the container it is in.



Practice

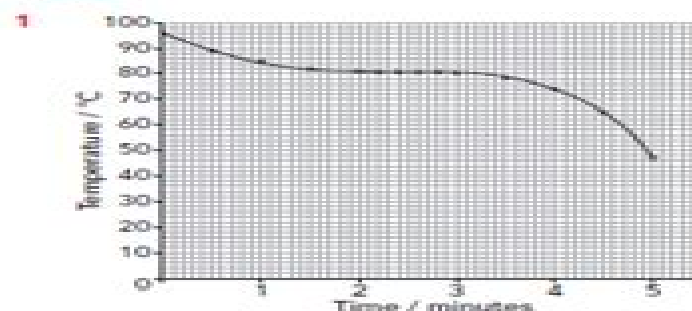
- 3 a B
b C
c A
d D
- 4 A – freezing (solidification); B – melting;
C – condensation; D – evaporation or boiling

Challenge

- 5 a radon
b radon and nitrogen
c nitrogen
d cobalt
e The sample of ethanoic acid is impure. The presence of impurities raises the boiling point of a substance.

Exercise 1.2

Focus



- 2 The student would need to use an oil bath (instead of the water bath) so she could heat the substance to a higher temperature.
- 3 The substance is freezing (solidifying) / turning from liquid to solid.

Practice

- 4 The temperature stays constant because energy is released as the substance solidifies / the molecules are giving out heat as they stop moving from place to place and become organised in a structured lattice arrangement / in a solid the molecules can only vibrate about fixed points / the heat released by the formation of new interactions (forces) between the particles keeps the temperature constant until all the substance is solid.

Chemistry Workbook Answers

Daniel C. Tofan



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