

Name \_\_\_\_\_

answer key

Date \_\_\_\_\_

# Stoichiometry: Mass - Mole Worksheet

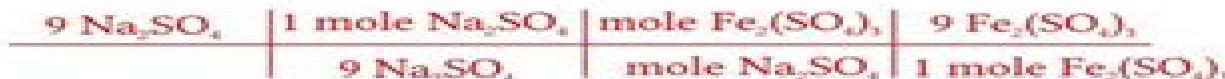
Based on the following chemical equation, please answer the following questions!



Calculate the amount of moles of NaCl needed to produce 75 g of FeCl<sub>3</sub>.



Calculate the amount of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> in grams produce 215 g Na<sub>2</sub>SO<sub>4</sub>.



Calculate the amount in grams of FeCl<sub>3</sub> produced with 95g of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>.



Calculate the amount of moles of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> needed to react with 123.5 g of NaCl



Based on the following chemical equation, please answer the following questions!



# Chemistry Study Guide The Mole Answers

**Paul B. Kelter**

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