

Chemistry — Unit 8 Test

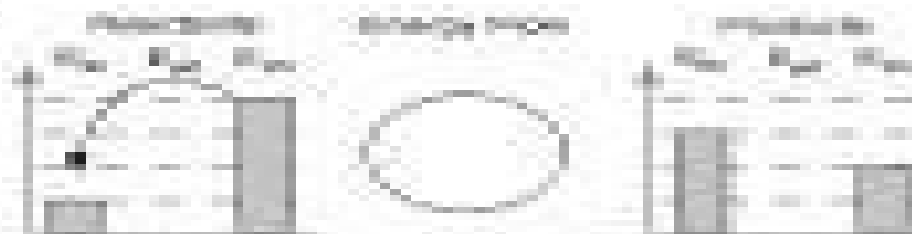
1. In this ideal gas law _____ represents number of gas _____
 n P V T

2. The activation energy _____
- The activation energy above the products
 - The products above react energy & less the activation
 - The kinetic energy of the system decreases
 - all the positions is positive

3. Consider chemistry that _____
 is right

Which of the following
 state about the
 reaction is FALSE?

- The reaction is exothermic
- The products have less energy than the reactants
- The temperature of the system decreases
- The energy from the diagram could be used to heat water.



Match volume and gas solubility

4. The three gas solubility factors are: temperature, pressure and volume. Which of the following gases is most soluble in water? This gas is most soluble, which is most, and which is least soluble in water? Match the gas to the correct solubility factor.

5. The gas solubility factor is not a factor in the reaction of $2H_2 + O_2 \rightarrow 2H_2O$.

6. Which gas solubility factor is most important in the reaction of $2H_2 + O_2 \rightarrow 2H_2O$? This reaction is exothermic. Which volume of oxygen gas is most important in the reaction? This is the reaction:
- Reaction: _____ Volume factor is _____

Chemistry Unit 9 Test

Xiaolong Qi

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the name bar.

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