

# COMMUNARDES!

LES ÉLÉPHANTS ROUGES



VENTS D'OUEST

# Communardes Lphants Rouges Wilfrid Lupano

**Wilfrid Lupano, Lucy Mazel**



### **Communardes Lphants Rouges Wilfrid Lupano:**

*Communardes ! - Les Eléphants rouges* Wilfrid Lupano, Lucy Mazel, 2015-09-30 On m appellera la dame aux l phants  
Hiver 1870 Pr lude de la Commune Alors que Paris assi g e par l arm e prussienne subit le froid et la famine Victorine onze  
ans passe le plus clair de son temps s occuper de Castor et Pollux les deux l phants du Jardin des plantes Cette passion pour  
les pachydermes a le don d nerver sa m re engag e dans le mouvement des femmes qui veulent s impliquer dans la d fense de  
la ville Mais Victorine est bourr e d imagination et elle veut tre la hauteur des ambitions de sa m re Nourrie par les exploits  
des c l bres l phants d Hannibal elle labore un plan pour lib rer Paris Un plan g nial d mesur contre lequel Bismarck ne peut  
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web in the case of a welcoming ceremony the welcome speech is considered to be one of the most important parts of the whole gathering in fact it is the most crucial aspect of the entire event this is because the welcome speech serves several purposes first it introduces the guests to the host

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web 1 define each remember oil rig oxidation is loss of e reduction is gain of e a oxidation b reduction c oxidizing agent d reducing agent 2 write half reactions for each of the following atoms or ions label each as oxidation or reduction use the activity series chart a al b ba2 c br 2 d ca e ga3 f h 2 g h

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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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web an oxidation reduction redox reaction is one in which one or more atoms change oxidation numbers oxidation occurs when an atom s oxidation state becomes more positive indicating that electrons have been lost reduction occurs when an atom s oxidation state becomes less positive indicating that electrons have been gained

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a helpful way to remember this

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23  $\text{Fe}^{2+} + \text{Co} \rightarrow \text{Co}^{2+} + \text{Fe}$   
oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^-$  reduction  $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$  24 3  $\text{Ag} + \text{Ni} \rightarrow \text{Ni}^{3+} + 3\text{Ag}$  oxidation reduction 25  $\text{Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$  oxidation reduction 26  $\text{O}_2$

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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent a  $\text{I}_2\text{O}_5$   
 $5\text{CO} + \text{I}_2\text{O}_5 \rightarrow 5\text{CO}_2 + \text{I}_2$   $5\text{C} + \text{O}_2 \rightarrow 5\text{CO}_2$

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation state and oxidation number concepts to identify redox reactions and to keep track of electrons

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web jun 28 2010 answers i determine what is oxidized and what in each reaction identify the oxidizing agent and the reducing agent also  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$   $\text{Sr}^0$  to  $\text{Sr}^{2+}$  oxidized reducing agent  $\text{O}_2$  to  $\text{O}^{2-}$  reduced ox ag  $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$   $\text{Li}^0$  to  $\text{Li}^{+1}$  oxidized red ag  $\text{S}^0$  to  $\text{S}^{2-}$  reduced ox ag  $3\text{Cs} + \text{Br}_2$

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cold war and even the personal dramas of a small town grammar school are shadowed by the threat of the nuclear arms race  
kermani s novel manages to capture these social tensions without sacrificing any of the all consuming