

Chemistry — Unit 8 Test

1. The following graph shows the relationship between the number of atoms and the mass of an element.

- a. The two axes for this graph are:
- b. The following statement is true concerning this graph:
 - i. The positive slope means increasing the mass does not increase the number of atoms.
 - ii. The negative slope means increasing the mass does not increase the number of atoms.
 - iii. The horizontal slope means the number of atoms does not change.
 - iv. The vertical slope means the number of atoms does not change.

2. The following diagrams show three different ways to represent the same element.

- a. The following statement is true concerning the three diagrams:
 - i. The first diagram is a Bohr model.
 - ii. The second diagram is a Lewis dot structure.
 - iii. The third diagram is a Venn diagram.
 - iv. The fourth diagram is a molecular model.
- b. The following statement is true concerning the three diagrams:
 - i. The first diagram shows the number of valence electrons.
 - ii. The second diagram shows the number of valence electrons.
 - iii. The third diagram shows the number of valence electrons.
 - iv. The fourth diagram shows the number of valence electrons.



Relationships between atomic structure and periodicity

- a. The following statement about atoms would not be true concerning the following statement:
 - i. The number of protons in an atom is the same as the number of neutrons.
 - ii. The number of protons in an atom is the same as the number of electrons.
 - iii. The number of protons in an atom is the same as the number of neutrons.
 - iv. The number of protons in an atom is the same as the number of neutrons.

b. The following statement about atoms would not be true concerning the following statement:

- i. The number of protons in an atom is the same as the number of neutrons.
- ii. The number of protons in an atom is the same as the number of neutrons.
- iii. The number of protons in an atom is the same as the number of neutrons.
- iv. The number of protons in an atom is the same as the number of neutrons.

Chemistry Unit 9 Test

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