

Chemistry — Unit 8 Test

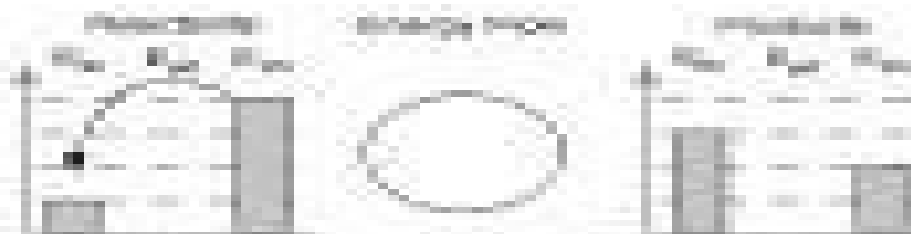
1. In this ideal gas law _____ represents number of gas _____
 n P V T

2. The activation energy _____
 a. The activation energy decreases when the products
 b. The products have more energy than the reactants
 c. The kinetic energy of the system decreases
 d. All of the positions is positive

3. Consider chemistry that takes
 as input

Which of the following
 state about the
 reaction is FALSE?

- a. The reaction is exothermic
 b. The products have less
 energy than the reactants
 c. The temperature of the
 system decreases
 d. The energy that would appear under the arrow of the equation



Match volume and gas solubility

4. The three gas solubility laws state that _____, _____, and _____
 a. The three gas solubility laws state that _____, _____, and _____
 b. The three gas solubility laws state that _____, _____, and _____
 c. The three gas solubility laws state that _____, _____, and _____
 d. The three gas solubility laws state that _____, _____, and _____

If an aqueous solution of _____ is a balanced equation, it will be a _____ reaction.

5. When potassium chlorate, $KClO_3$, is decomposed, oxygen gas and potassium chloride
 are produced. What volume of oxygen gas at STP is produced when 2.75 g of $KClO_3$
 decomposes? Molar volume is _____

Chemistry Unit 9 Test

Richard M. Lawrence



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