

Intro. to Chemical Reactions

Electrolyte Reactions

1. Write down formulas for the following compounds:

a) Barium hydroxide $\text{Ba}(\text{OH})_2$

b) Sodium phosphate Na_3PO_4

c) Magnesium phosphate $\text{Mg}_3(\text{PO}_4)_2$

d) Calcium iodide CaI_2

2. Write the formulas for the following acids:

a) HClO₄ Hydrogen perchlorate

b) H₃PO₄ Hydrogen phosphate

c) H₂SO₄ Hydrogen sulfate

d) H₂CO₃ Hydrogen carbonate

Electrolyte Reactions: Physical Theory Review

There is often a chemical reaction that does not go forward:



There are reasons that you can name as reasons why things don't go:

Common/Shared reagents with **shared partial and partial charges** and **partial bonds**

There is often no clear relationship between the charges and the products but there are other ways to do it:

Common/Shared Charges

1. Given the following reaction, write the equation that goes along with it:



Common/Shared charges with either positive and negative charges makes for a strong reaction

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