



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the steps for naming the compound)



Steps for Naming:

1. Identify the cation (metal) and anion (non-metal).
2. Name the cation first, followed by the anion.
3. Use the appropriate suffix for the anion (usually -ide).
4. If the cation is a transition metal, include its oxidation state in parentheses.
5. If the anion is a polyatomic ion, use its name.
6. If the compound is a binary ionic compound, use the prefix system for the anion.

Binary Compounds

Formula	Name
NaCl	Sodium Chloride
CaO	Calcium Oxide
Fe ₂ O ₃	Iron(III) Oxide
Al ₂ Cl ₆	Aluminum Chloride
CO ₂	Carbon Dioxide
SO ₂	Sulfur Dioxide
NO ₂	Nitrogen Dioxide
H ₂ O	Water
CH ₄	Methane
NH ₃	Nitrogen
HCl	Hydrogen Chloride

Common Ions and Polyatomic Ions

Formula	Name
Na ⁺	Sodium
Ca ²⁺	Calcium
Fe ²⁺	Iron(II)
Fe ³⁺	Iron(III)
Al ³⁺	Aluminum
CO ₃ ²⁻	Carbonate
SO ₄ ²⁻	Sulfate
NO ₃ ⁻	Nitrate
H ₂ O	Water
CH ₄	Methane
NH ₃	Nitrogen
HCl	Hydrogen Chloride

Naming Compounds with a Central and Peripheral



Carbon Dioxide

Steps for Naming:

1. Identify the central atom and the peripheral atom.
2. Name the central atom first, followed by the peripheral atom.
3. Use the appropriate suffix for the peripheral atom (usually -ide).
4. If the central atom is a transition metal, include its oxidation state in parentheses.
5. If the peripheral atom is a polyatomic ion, use its name.

Naming Practice

Formula	Name	Formula	Name
NaCl	Sodium Chloride	CaO	Calcium Oxide
Fe ₂ O ₃	Iron(III) Oxide	Al ₂ Cl ₆	Aluminum Chloride
CO ₂	Carbon Dioxide	SO ₂	Sulfur Dioxide
NO ₂	Nitrogen Dioxide	H ₂ O	Water
CH ₄	Methane	NH ₃	Nitrogen
HCl	Hydrogen Chloride		

Common Names for Specific Compounds

Formula	Common Name	Formula	Common Name
NaCl	Salt	CaO	Quicklime
Fe ₂ O ₃	Rust	Al ₂ Cl ₆	Aluminum Chloride
CO ₂	Carbon Dioxide	SO ₂	Sulfur Dioxide
NO ₂	Nitrogen Dioxide		

Cheat Sheet For Naming Compounds

Robert Sidney Cahn



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Naming Chemical Compounds, 2000 **The Etymology of Chemical Names**, 2019-10-08 Etymology of Chemical Names gives an overview of the development of the current chemical nomenclature tracing its sources and changing rules as chemistry progressed over the years This book is devoted to provide a coherent picture how the trivial and systematic names shall be used and how the current IUPAC rules help to reconcile the conflicting demands **The 100 Most Important Chemical Compounds** Richard L. Myers, 2007-08-30 What is a chemical compound Compounds are substances that are two or more elements combined together chemically in a standard proportion by weight Compounds are all around us they include familiar things such as water and more esoteric substances such as triuranium octaoxide the most commonly occurring natural source for uranium This reference guide gives us a tour of 100 of the most important common unusual and intriguing compounds known to science Each entry gives an extensive explanation of the composition molecular formula and chemical properties of the compound In addition each entry reviews the relevant chemistry history and uses of the compound

with discussions of the origin of the compound's name, the discovery or first synthesis of the compound, production statistics and uses of the compound. *The Naming and Indexing of Chemical Compounds* Chemical Abstracts (Periodical), 1962

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Table of Contents Cheat Sheet For Naming Compounds

1. Understanding the eBook Cheat Sheet For Naming Compounds
 - The Rise of Digital Reading Cheat Sheet For Naming Compounds
 - Advantages of eBooks Over Traditional Books
2. Identifying Cheat Sheet For Naming Compounds
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Cheat Sheet For Naming Compounds
 - User-Friendly Interface
4. Exploring eBook Recommendations from Cheat Sheet For Naming Compounds

- Personalized Recommendations
 - Cheat Sheet For Naming Compounds User Reviews and Ratings
 - Cheat Sheet For Naming Compounds and Bestseller Lists
5. Accessing Cheat Sheet For Naming Compounds Free and Paid eBooks
 - Cheat Sheet For Naming Compounds Public Domain eBooks
 - Cheat Sheet For Naming Compounds eBook Subscription Services
 - Cheat Sheet For Naming Compounds Budget-Friendly Options
 6. Navigating Cheat Sheet For Naming Compounds eBook Formats
 - ePub, PDF, MOBI, and More
 - Cheat Sheet For Naming Compounds Compatibility with Devices
 - Cheat Sheet For Naming Compounds Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Cheat Sheet For Naming Compounds
 - Highlighting and Note-Taking Cheat Sheet For Naming Compounds
 - Interactive Elements Cheat Sheet For Naming Compounds
 8. Staying Engaged with Cheat Sheet For Naming Compounds
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Cheat Sheet For Naming Compounds
 9. Balancing eBooks and Physical Books Cheat Sheet For Naming Compounds
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Cheat Sheet For Naming Compounds
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Cheat Sheet For Naming Compounds
 - Setting Reading Goals Cheat Sheet For Naming Compounds
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Cheat Sheet For Naming Compounds

- Fact-Checking eBook Content of Cheat Sheet For Naming Compounds
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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