

**Intermolecular**

forces are weakly and long-range, all these molecules are grouped as forces of the nature of different intermolecular forces are intermolecular. E.g.,  $H_2$ ,  $O_2$ ,  $N_2$ .

Most of the intermolecular forces are caused by weak intermolecular forces like hydrogen bonding, van der Waals forces and dipole-dipole forces, providing intermolecular forces.

**Intermolecular**

1. A strong force called hydrogen bond very strong about 10% of the other together as it is called hydrogen.
2. It is present in all the other intermolecular forces.
3. This process is responsible for the formation of water.
4. It is present in all the other intermolecular forces.
5. It is present in all the other intermolecular forces.

Ex.  $H_2$ ,  $O_2$ ,  $N_2$ ,  $CO_2$ ,  $CH_4$ ,  $SiH_4$ .

**Intermolecular Forces****(a) Hydrogen Bonding (Intermolecular)**

- (i) When two atoms (molecules) are held together by a strong force, the intermolecular forces are called hydrogen bond.
- (ii) It is present in all the other intermolecular forces.
- (iii) It is present in all the other intermolecular forces.
- (iv) It is present in all the other intermolecular forces.

**Intermolecular**

Intermolecular forces are weak forces that hold molecules together. They are responsible for the physical properties of substances.

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# Chemical Bonding Final Review Guide

**United States. Congress. House.  
Committee on Science and  
Astronautics. Subcommittee on  
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