

Name: \_\_\_\_\_ Block: \_\_\_\_\_

## **Another Balancing Equations Sheet!**

- 1) \_\_\_\_\_  $\text{AlBr}_3 + \text{K} \rightarrow \text{KBr} + \text{Al}$
- 2) \_\_\_\_\_  $\text{FeO} + \text{PdF}_2 \rightarrow \text{FeF}_2 + \text{PdO}$
- 3) \_\_\_\_\_  $\text{P}_4 + \text{Br}_2 \rightarrow \text{PBr}_3$
- 4) \_\_\_\_\_  $\text{LiCl} + \text{Br}_2 \rightarrow \text{LiBr} + \text{Cl}_2$
- 5) \_\_\_\_\_  $\text{PbBr}_2 + \text{HCl} \rightarrow \text{HBr} + \text{PbCl}_2$
- 6) \_\_\_\_\_  $\text{CoBr}_3 + \text{CaSO}_4 \rightarrow \text{CaBr}_2 + \text{Co}_2(\text{SO}_4)_3$
- 7) \_\_\_\_\_  $\text{Na}_3\text{P} + \text{CaF}_2 \rightarrow \text{NaF} + \text{Ca}_3\text{P}_2$
- 8) \_\_\_\_\_  $\text{Mn} + \text{HI} \rightarrow \text{H}_2 + \text{MnI}_3$
- 9) \_\_\_\_\_  $\text{Li}_3\text{PO}_4 + \text{NaBr} \rightarrow \text{Na}_3\text{PO}_4 + \text{LiBr}$
- 10) \_\_\_\_\_  $\text{CaF}_2 + \text{Li}_2\text{SO}_4 \rightarrow \text{CaSO}_4 + \text{LiF}$
- 11) \_\_\_\_\_  $\text{HBr} + \text{Mg}(\text{OH})_2 \rightarrow \text{MgBr}_2 + \text{H}_2\text{O}$
- 12) \_\_\_\_\_  $\text{LiNO}_3 + \text{CaBr}_2 \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{LiBr}$
- 13) \_\_\_\_\_  $\text{AgNO}_3 + \text{Li} \rightarrow \text{LiNO}_3 + \text{Ag}$
- 14) \_\_\_\_\_  $\text{Si}(\text{OH})_4 + \text{NaBr} \rightarrow \text{SiBr}_4 + \text{NaOH}$
- 15) \_\_\_\_\_  $\text{NaCN} + \text{CuCO}_3 \rightarrow \text{Na}_2\text{CO}_3 + \text{Cu}(\text{CN})_2$
- 16) \_\_\_\_\_  $\text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow (\text{NH}_4)_2\text{SO}_4$

# Chemical Equations Study Guide With Answers

**Thomas Elliott Taylor**

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