

## Year 7 Chemistry: Chemical Reactions

### 1. What is a Chemical Reaction?

- A chemical reaction is a process where substances (reactants) change into new substances (products).
- It involves breaking old bonds and forming new ones.
- **Example:** Burning wood, rusting of iron, baking a cake.

### 2. Signs of a Chemical Reaction

You can tell a chemical reaction has happened by looking for these signs:

- **Color change** (e.g., iron rusting turns orange-brown).
- **Gas production** (e.g., bubbles forming when vinegar reacts with baking soda).
- **Temperature change** (heat may be given off or absorbed).
- **Formation of a precipitate** (a solid appears in a liquid).
- **Light or sound production** (e.g., fireworks).

### 3. What Happens During a Chemical Reaction?

- Atoms rearrange to form new substances.
- The total number of atoms stays the same (Law of Conservation of Mass).
- **Example:** Hydrogen gas + Oxygen gas  $\rightarrow$  Water ( $H_2O$ ).

### 4. Precipitate Formation

- A precipitate is a solid that forms from a solution during a reaction.
- **Example:** Mixing silver nitrate and sodium chloride forms a white precipitate of silver chloride.

### 5. Using the Particle Model to Explain Chemical Reactions

- Matter is made up of tiny particles (atoms and molecules).
- During a reaction, particles collide, break apart, and form new substances.
- Temperature and concentration affect how fast particles react.

### 6. Making Predictions in Science

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**Dorothy A. McGinniss**



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