

Year 7 Chemistry: Chemical Reactions

1. What is a Chemical Reaction?

- A chemical reaction is a process where substances (reactants) change into new substances (products).
- It involves breaking old bonds and forming new ones.
- Example: Burning wood, rusting of iron, baking a cake.

2. Signs of a Chemical Reaction

You can tell a chemical reaction has happened by looking for these signs:

- Color change (e.g., iron rusting turns orange-brown).
- Gas production (e.g., bubbles forming when vinegar reacts with baking soda).
- Temperature change (heat may be given off or absorbed).
- Formation of a precipitate (a solid appears in a liquid).
- Light or sound production (e.g., fireworks).

3. What Happens During a Chemical Reaction?

- Atoms rearrange to form new substances.
- The total number of atoms stays the same (Law of Conservation of Mass).
- Example: Hydrogen gas + Oxygen gas \rightarrow Water (H_2O).

4. Precipitate Formation

- A precipitate is a solid that forms from a solution during a reaction.
- Example: Mixing silver nitrate and sodium chloride forms a white precipitate of silver chloride.

5. Using the Particle Model to Explain Chemical Reactions

- Matter is made up of tiny particles (atoms and molecules).
- During a reaction, particles collide, break apart, and form new substances.
- Temperature and concentration affect how particles react.

6. Making Predictions in Science

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