

Name _____

Chemistry 1411 Practice Exam 1, Chapters 1-3

Show your calculations clearly on show work problems. A periodic table is also included.
Good luck!

1. Give the number of significant figures in the following numbers (3 pts):

1) 6.022×10^{23} atoms/mol _____

2) 0.049001 g _____

a) 3, 5

b) 4, 2

c) 23, 7

d) 4, 5

2. A rectangular block with a length, width, and height of 2.55 cm, 2.55 cm, and 1.44 cm has a density of 0.850 g/cm³. What is the mass of this object in grams? (3 pts)

a) 5.46 g

b) 7.96 g

c) 8.65 g

d) 10.4 g

3. (3 pts) Isotopes are atoms that:

a) have different atomic numbers

b) have the same mass number

c) have the same number of protons but different numbers of neutrons

d) have different numbers of electrons

4. A general summary based on observed and measurable behavior in many different systems is called a: (3 pts)

a) hypothesis

b) natural law

c) model

d) theory

5. What is the number of protons, neutrons, and electrons in the iron-56 atom? (3 pts)

a) 56 protons; 26 neutrons; 24 electrons

b) 26 protons; 30 neutrons; 24 electrons

c) 24 protons; 56 neutrons; 26 electrons

d) 26 protons; 30 neutrons; 26 electrons

6. Which of the following pairs of elements is a transition element and a halogen? (3 pts)

a) W and Si

b) Cr and Ar

c) Au and Br

d) Ba and U

7. Which of the following compounds is molecular? (3 pts)

a) BaCl₂

b) NaOH

c) SF₆

d) KNO₃

8. Which of the following elements is a noble gas? (3 pts)

a) Mg

b) Ne

c) Pb

d) Br

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