

Name _____

Chemistry 1411 Practice Exam 1, Chapters 1-3

Show your calculations clearly on show work problems. A periodic table is also included.
Good luck!

1. Give the number of significant figures in the following numbers (3 pts):
1) 6.022×10^{23} atoms/mol _____ 2) 0.049001 g _____
a) 3, 5 b) 4, 2 c) 23, 7 d) 4, 5
2. A rectangular block with a length, width, and height of 2.55 cm, 2.55 cm, and 1.44 cm has a density of 0.850 g/cm³. What is the mass of this object in grams? (3 pts)
a) 5.46 g b) 7.96 g c) 8.65 g d) 10.4 g
3. (3 pts) Isotopes are atoms that:
a) have different atomic numbers
b) have the same mass number
c) have the same number of protons but different numbers of neutrons
d) have different numbers of electrons
4. A general summary based on observed and measurable behavior in many different systems is called a: (3 pts)
a) hypothesis b) natural law c) model d) theory
5. What is the number of protons, neutrons, and electrons in the iron-56 atom? (3 pts)
a) 56 protons; 26 neutrons; 24 electrons b) 26 protons; 30 neutrons; 24 electrons
c) 24 protons; 56 neutrons; 26 electrons d) 26 protons; 30 neutrons; 26 electrons
6. Which of the following pairs of elements is a transition element and a halogen? (3 pts)
a) W and Si b) Cr and Ar c) Au and Br d) Ba and U
7. Which of the following compounds is molecular? (3 pts)
a) BaCl₂ b) NaOH c) SF₆ d) KNO₃
8. Which of the following elements is a noble gas? (3 pts)
a) Mg b) Xe c) Pb d) Br

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