

1. In the list below, circle the properties of acids and box the properties of bases.

Sour taste

Feel slippery

Contains hydroxide ions

Turns litmus blue

Reacts with metals to form H₂

Turns litmus red

Bitter taste

Contains hydrogen ions

Corrosive

2. Complete the chart below:

Acid-Base Definitions

Type	Acid Definition	Base Definition
Arrhenius	Donates an H ⁺ ion in solution	Donates an OH ⁻ ion in solution
Bronsted-Lowry	Donates an H ⁺ ion	Accepts an H ⁺ ion
Lewis	Accepts an electron pair	Donates an electron pair

3. What term(s) describe a substance that can act as an acid or a base? **Amphoteric or amphiprotic**

4. Neutralization reactions always produce a **SALT** and **WATER**

5. If the [H⁺] is greater than the [OH⁻], the solution is

acidic basic neutral

6. If the [OH⁻] is greater than the [H⁺], the solution is

acidic basic neutral

7. If the [H⁺] is equal to the [OH⁻], the solution is

acidic basic neutral

8. If the pH is 7, the solution is

acidic basic neutral

9. If the pH is greater than 7, the solution is

acidic basic neutral

10. If the pH is less than 7, the solution is

acidic basic neutral

11. How many moles of lithium hydroxide are required to neutralize 19.2 grams of chloric acid?



$$(19.2 \text{ g HClO}_3) \times (1 \text{ mol HClO}_3 / 84 \text{ g HClO}_3) \times (1 \text{ mol LiOH} / 1 \text{ mol HClO}_3) = 0.228 \text{ mol LiOH}$$

12. How many grams of calcium hydroxide are required to neutralize 132.4 grams of nitric acid?



$$(132.4 \text{ g HNO}_3) \times (1 \text{ mol HNO}_3 / 63 \text{ g HNO}_3) \times (1 \text{ mol Ca(OH)}_2 / 2 \text{ mol HNO}_3) \times (74 \text{ g Ca(OH)}_2 / 1 \text{ mol Ca(OH)}_2) = 77.76 \text{ g Ca(OH)}_2$$

Chemistry Chapter 19 Acids Bases Study Guide Answers

Iris Stovall,Donald J. DeCoste

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