

**Matching**

Match each item with the correct statement below.

- a. molar volume
- b. molar mass
- c. atomic mass

- 1. the number of grams of an element that is numerically equal to the atomic mass of the element in amu
- 2. the mass of a mole of any element or compound
- 3. the volume occupied by a mole of any gas at STP

Match each item with the correct statement below.

- a. representative particle
- b. mole
- c. Avogadro's number
- d. percent composition
- e. standard temperature and pressure
- f. empirical formula

- 4. the number of representative particles of a substance present in 1 mole of that substance
- 5. an atom, an ion, or a molecule, depending upon the way a substance commonly exists
- 6. the SI unit used to measure amount of substance
- 7. OPM and 1 atm
- 8. the percent by mass of each element in a compound
- 9. the smallest whole number ratio of the atoms in a compound

**Multiple Choice**

Identify the choice that best completes the statement or answers the question.

- 1. What SI unit is used to measure the number of representative particles in a substance?  
a. kilogram      b. ampere      c. kelvin      d. mole
- 2. How many hydrogen atoms are in 5 molecules of isopropyl alcohol,  $C_3H_8O$ ?  
a.  $5 \times (6.02 \times 10^{23})$       c. 35  
b. 5      d.  $35 \times (6.02 \times 10^{23})$
- 3. If a dozen Clementine oranges have a mass of 744 g, what will be the mass of 15 Clementines?  
a. 11.16 kg      b. 595 g      c. 896 g      d. 930 g
- 4. Which of the following elements exists as a diatomic molecule?  
a. neon      b. lithium      c. nitrogen      d. sulfur
- 5. Avogadro's number of representative particles is equal to one \_\_\_\_\_.  
a. kilogram      b. gram      c. kelvin      d. mole
- 6. Which of the following is equal to Avogadro's number?  
a. the number of atoms of bromine in 1 mol  $Br_2$   
b. the number of molecules of chlorine in 0.50 mol  $Cl_2$   
c. the number of molecules of nitrogen in 1 mol  $N_2$   
d. the number of atoms in 1 mol CO

# Chemistry Chemical Quantities Study Guide Answers

**Iris Stovall**

## Chemistry Chemical Quantities Study Guide Answers:

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this manual will provide you with opportunities to sharpen your skills and evaluate your comprehension of the material in the texts The worked out solutions at the end of this manual walk you step by step through the methods of arriving at the answer to those same problems which have an answer only in the answer key Used in conjunction with your textbook and classroom lecture notes this book offers an essential learning opportunity to the chemistry student Back cover

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