

Chemistry Module 3 – Water

Outcome 1: Water is distribute on Earth as a solid, liquid and gas.

Outcome 1.1: Define the term solute, solvent and solution

Solvents: Liquid in which different substances dissolve to form homogeneous mixtures or solutions. Eg. Water, ethanol and milk are solvents.

Solute: A substance present in the smaller amount) that is dissolved in a solvent to produce a solution. Usually a solid. Eg. salt, sugar.

Solution: A homogenous mixture of two or more substances (usually a solid and a liquid). Solutions may exist as solids, liquids or gases.

Outcome 1.2: Identify the importance of water as a solvent

Importance of water:

- Is a solvent for all the chemicals in a living cell and the medium in which all the chemical reactions occur.
- Oxygen and carbon dioxide dissolve in water in the bloodstream and are then carried to different parts of the body.
- Various nitrogenous waste products of metabolism (e.g. urea) can dissolve in water and be eliminated from the body as urine.
- Carbon dioxide dissolves in the cellular water of leaf cells and is then available as a raw material for photosynthesis.
- Solutions of acids in water are used by our body to digest food.
- Hormones produced by our glands can be transported to various organs because they are soluble in the aqueous body fluids.
- Oceans and rivers dissolve atmospheric oxygen, which makes it available to aquatic organisms.
- A reactant or product in many biological reactions such as photosynthesis and cellular respiration.

Outcome 1.3: Compare the state, percentage and distribution of water in the biosphere, lithosphere, hydrosphere and atmosphere.

Sphere of the Earth	Classification according to physical state	% of sphere that is water
Hydrosphere	Solid: polar ice, glaciers Liquid: Oceans, inland seas, lakes, rivers	95-99%
Biosphere	Liquid: Cell protoplasm, extracellular water, transport systems. Gas: Water vapour in intercellular and some organ spaces.	60-95%
Lithosphere	Solid: Frozen soil, hydrated minerals.	Less than 10%

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