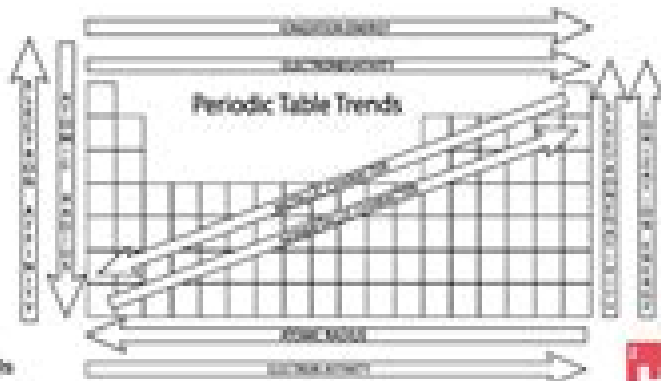


PERIODIC TABLE OF THE ELEMENTS

Atomic Number → **1** ← Symbol
Name → **Hydrogen** ← Atomic Weight

-  Solid
 Liquid
 Gas
- *At room temperature



Group 1 →

Period ↓

1 H

2 Li Be

3 Na Mg

4 K Ca Sc Ti V Cr Mn Fe Co Ni Cu Zn Ga Ge As Se Br Kr

5 Rb Sr Y Zr Nb Mo Tc Ru Rh Pd Ag Cd In Sn Sb Te I Xe

6 Cs Ba 57-71 Hf Ta W Re Os Ir Pt Au Hg Tl Pb Bi Po At Rn

7 Fr Ra 89-103 Rf Db Sg Bh Hs Mt Ds Rg Cn Nh Fl Mc Lv Ts Og

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He

Ne

Ar

Kr

Xe

Rn

Og

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Electron Configuration Effects

The diagram illustrates the distribution of atomic orbitals across the periodic table. The s-block (purple) includes groups 1 and 2. The p-block (orange) includes groups 13 through 18. The d-block (green) includes groups 3 through 10. The f-block (light green) includes the lanthanide and actinide series. The orbitals are labeled as follows:

- s:** Purple blocks in groups 1 and 2.
- p:** Orange blocks in groups 13 through 18.
- d:** Green blocks in groups 3 through 10.
- f:** Light green blocks in the lanthanide and actinide series.

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Hart Mcdougal



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