

Chemistry

BASIC CONCEPTS

- The percentage of oxygen in NaOH is
 - 40
 - 60
 - 8
 - 10
- What quantity of limestone (CaCO_3) on heating will give 56 kg of CaO ?
 - 1000 kg
 - 56 kg
 - 44 kg
 - 100 kg
- Which of the following has least mass?
 - 2 gram atom of nitrogen
 - 3×10^{23} atoms of C
 - 1 mole of S
 - 7.0 g of Ag.
- 1 mole of CH_4 contains
 - 6.02×10^{23} atoms of H
 - 4 g-atom of hydrogen
 - 1.81×10^{23} molecules of CH_4
 - 3.0 g of carbon
- How many moles of Helium gas occupy 22.4 L at 0°C at 1 atm. Pressure?
 - 0.11
 - 0.90
 - 1.0
 - 1.11
- The number of atoms contained in 11.2 L of SO_2 at S.T.P. are
 - $3/2 \times 6.02 \times 10^{23}$
 - $2 \times 6.02 \times 10^{23}$
 - 6.02×10^{23}
 - $4 \times 6.02 \times 10^{23}$
- The Number of oxygen atoms in 4.4 g of CO_2 is approximately
 - 1.2 $\times 10^{23}$
 - 6×10^{22}
 - 6×10^{23}
 - 12×10^{22}
- The total number of protons in 10 g of calcium carbonate is ($N_A = 6.023 \times 10^{23}$)
 - 1.5057×10^{24}
 - 2.0478×10^{24}
 - 3.0115×10^{24}
 - 4.0956×10^{24}
- 2 g of oxygen contains number of atoms equal to that in
 - 0.5 g of hydrogen
 - 4 g of sulphur
 - 7 g of nitrogen
 - 2.3 g of sodium
- If N_A is Avogadro's number, then number of valence electrons in 4.2 g of nitride ions N^{3-} is
 - $2.4 N_A$
 - $4.2 N_A$
 - $1.6 N_A$
 - $3.2 N_A$

ANSWERS

1	A	6	A
2	D	7	A
3	B	8	C
4	B	9	B
5	C	10	A

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Table of Contents Chemistry Mcqs For Entry Test Of Engineering

1. Understanding the eBook Chemistry Mcqs For Entry Test Of Engineering
 - The Rise of Digital Reading Chemistry Mcqs For Entry Test Of Engineering
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Mcqs For Entry Test Of Engineering
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Mcqs For Entry Test Of Engineering
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Mcqs For Entry Test Of Engineering

- Personalized Recommendations
- Chemistry Mcqs For Entry Test Of Engineering User Reviews and Ratings
- Chemistry Mcqs For Entry Test Of Engineering and Bestseller Lists

5. Accessing Chemistry Mcqs For Entry Test Of Engineering Free and Paid eBooks

- Chemistry Mcqs For Entry Test Of Engineering Public Domain eBooks
- Chemistry Mcqs For Entry Test Of Engineering eBook Subscription Services
- Chemistry Mcqs For Entry Test Of Engineering Budget-Friendly Options

6. Navigating Chemistry Mcqs For Entry Test Of Engineering eBook Formats

- ePUB, PDF, MOBI, and More
- Chemistry Mcqs For Entry Test Of Engineering Compatibility with Devices
- Chemistry Mcqs For Entry Test Of Engineering Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Chemistry Mcqs For Entry Test Of Engineering
- Highlighting and Note-Taking Chemistry Mcqs For Entry Test Of Engineering
- Interactive Elements Chemistry Mcqs For Entry Test Of Engineering

8. Staying Engaged with Chemistry Mcqs For Entry Test Of Engineering

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chemistry Mcqs For Entry Test Of Engineering

9. Balancing eBooks and Physical Books Chemistry Mcqs For Entry Test Of Engineering

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Chemistry Mcqs For Entry Test Of Engineering

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Chemistry Mcqs For Entry Test Of Engineering

- Setting Reading Goals Chemistry Mcqs For Entry Test Of Engineering
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chemistry Mcqs For Entry Test Of Engineering

- Fact-Checking eBook Content of Chemistry Mcqs For Entry Test Of Engineering
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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