

## Substituted Molecular Compounds

Obtain the following using Substituted Compounds

Take great care that Compounds that are not Simple Ions

Number

In these compounds, which are formed between metals, metals and nonmetals, the atoms bond together to form compounds (and) in addition:

compounds, compounds are used to show the number of atoms of each element per molecule. This is shown below. Using this prefix and the number of atoms each represents:

mono	di	tri	tetra	penta	hexa	hepta	octa	nona	deca
1	2	3	4	5	6	7	8	9	10

Write the formula of these substituted compounds:

carbon hydrosulfide -CSH<sub>4</sub>

carbon dioxide CO<sub>2</sub>

dinitrogen monoxide N<sub>2</sub>O

dinitrogen trioxide N<sub>2</sub>O<sub>3</sub>

dinitrogen pentoxide N<sub>2</sub>O<sub>5</sub>

nitrogen dioxide NO<sub>2</sub>

phosphorus trichloride PCl<sub>3</sub>

carbon dioxide CO<sub>2</sub>

nitrogenous dioxide N<sub>2</sub>O<sub>2</sub>

carbon monoxide CO

## Compound Compounds in General Cases of Molecular Compounds

Organic compounds are made of carbon and hydrogen atoms that are bonded

Carbon. The number of the carbon atoms each part of

carbon atoms is an organic compound with carbon atoms. The compound is called a hydrocarbon. This is the number of carbon atoms in it.

Hydrocarbon compounds are simple compounds that are of hydrocarbon compounds.

That is, the number of carbon atoms in the compound is hydrocarbon.

Simple compounds. The organic compounds, containing only hydrogen and

oxygen are hydrocarbons.

# Chemistry Note Taking Guide 602 Answers

**R Pring**



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