

PRACTICE CHEMISTRY PLACEMENT EXAM

1. How many of the numbers below have 5 significant figures?

0.0054 19.000 0.00006 1.6090×10^8 13607

- a) 1
- b) 2
- c) 3
- d) 4
- e) 5

2. Nitric acid is a solution of which of the following dissolved in water?

- a) HNO_4
- b) H_2NO_3
- c) HNO_2
- d) HNO_3
- e) H_2NO

3. Which of the following is the smallest mass?

- a) 2.1 kg
- b) 4.2×10^{10} ng
- c) 5.8×10^2 g
- d) 8.4×10^4 cg
- e) 6.7×10^4 mg

4. If 250 mL of a 0.50 M NaCl solution is diluted to 840 mL, what is the molarity of the resulting solution?

- a) 0.15 M
- b) 6.7 M
- c) 0.60 M
- d) 1.7 M
- e) 0.0025 M

5. Iron(III) sulfite has the formula

- a) Fe_3SO_3
- b) $\text{Fe}_2(\text{SO}_4)_3$
- c) $\text{Fe}_2(\text{SO}_3)_3$
- d) Fe_2SO_4
- e) Fe_2SO_3

Chemistry Placement Test Practice

Donald E. Brook

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