

PRACTICE CHEMISTRY PLACEMENT EXAM

1. How many of the numbers below have 5 significant figures?

0.0054

19.000

0.00006

1.6090×10^5

13607

- a) 1
b) 2
c) 3
d) 4
e) 5
2. Nitric acid is a solution of which of the following dissolved in water?
- a) HNO_4
b) H_2NO_3
c) HNO_2
d) HNO_3
e) H_2NO
3. Which of the following is the smallest mass?
- a) 2.1 kg
b) 4.2×10^{10} ng
c) 5.8×10^2 g
d) 8.4×10^4 cg
e) 6.7×10^4 mg
4. If 250 mL of a 0.50 M NaCl solution is diluted to 840 mL, what is the molarity of the resulting solution?
- a) 0.15 M
b) 6.7 M
c) 0.60 M
d) 1.7 M
e) 0.0025 M
5. Iron(III) sulfite has the formula
- a) Fe_2SO_3
b) $\text{Fe}_2(\text{SO}_4)_3$
c) $\text{Fe}_2(\text{SO}_3)_3$
d) Fe_2SO_4
e) Fe_2SO_3

Chemistry Placement Test Practice

Donald E. Brook



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