

Chapter 9: Alcohols, Ethers, and Related Compounds

## Chapter Review

### General facts about $\text{ROH}$ , $\text{ROH}_2$ , and oxides

All three compounds contain an O atom that is  $sp^2$  hybridized and tetrahedral (9.2).



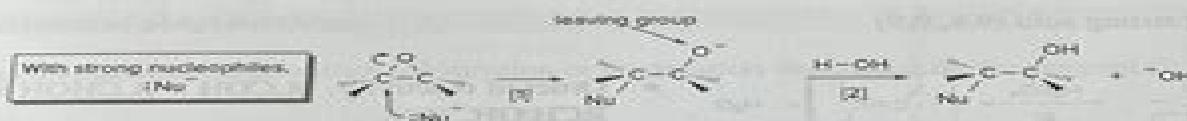
- All three compounds have polar C–O bonds, but only alcohols have an O–H bond for intermolecular hydrogen bonding (9.4).



- Alcohols and ethers do not contain a good leaving group. Nucleophilic substitution can occur only after the  $\text{OH}$  (or  $\text{OR}$ ) group is converted to a better leaving group ( $\text{O}^-$ ).



- Epoxides have a leaving group located in a strained three-membered ring, making them reactive to strong nucleophiles and acids  $\text{HZ}$  that contain a nucleophilic atom  $Z$  (9.16).



### A new reaction of carbocations (9.9)

- Less stable carbocations rearrange to more stable carbocations by shift of a hydrogen atom or an alkyl group. Besides rearrangement, carbocations also react with nucleophiles (7.12) and bases (8.6).



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**K. Peter C. Vollhardt, Neil E. Schore**

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