



CHEMISTRY

CHEAT SHEET



STATES OF MATTER

STATES OF MATTER CHEAT SHEET

Matter		Physical and Chemical Changes	
Matter is anything that takes up space and has volume. It has mass and is made of atoms and molecules.		Physical Change: A change in which the substance's physical properties change, but its chemical identity remains the same. Examples: melting, freezing, boiling, condensation, sublimation, deposition. Chemical Change: A change in which the substance's chemical identity changes. Examples: burning, rusting, digestion, photosynthesis.	
Phase	Description	Arrangement of Particles	Examples
SOLID	- Lowest energy state of matter. - Particles are closely packed in a regular, repeating pattern. - Highly ordered and rigid structure.		
LIQUID	- Intermediate state. - Particles are close together but can move past each other. - Disordered arrangement.		
GAS	- Highest energy state of matter. - Particles are far apart and move rapidly in all directions. - Disordered arrangement.		
PLASMA	- Highest energy state of matter. - Consists of free electrons and positive ions.		
Relative Energy Levels: The energy levels of matter are ordered from lowest to highest as follows: Solid < Liquid < Gas < Plasma.		Relative Energy Levels: Solid Liquid Gas Plasma	

Chemistry States Of Matter Study Guide

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