

Chemistry Study Guide

Periodic Table of the Elements

<div>1 H</div>																	<div>18 Ar</div>	<div>19 K</div>	<div>20 Ca</div>											<div>36 Kr</div>	<div>37 Rb</div>	<div>38 Sr</div>											<div>54 Xe</div>	<div>55 Cs</div>	<div>56 Ba</div>															<div>86 Rn</div>	<div>87 Fr</div>	<div>88 Ra</div>															<div>118 Og</div>																																
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Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎

Matter:

Anything that has **mass** and **occupies space** ✎

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** ✎

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

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Arshad Iqbal



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