

**CHEM 110**  
**Extra In-Class Practice Problems**  
**Lecture Set 3**

1. Answer the following questions dealing with Avogadro's number.

A. How many molecules present in 2.0 moles  $\text{CH}_4$ ?

$$\frac{2.0 \text{ mol } \text{CH}_4}{1 \text{ mol } \text{CH}_4} \times \frac{6.022 \times 10^{23} \text{ molecules}}{1 \text{ mol } \text{CH}_4} = 1.2 \times 10^{24} \text{ molecules } \text{CH}_4$$

How many atoms are present?

$$\frac{1.2 \times 10^{24} \text{ molecules } \text{CH}_4}{1 \text{ molecule } \text{CH}_4} \times \frac{5 \text{ atoms}}{1 \text{ molecule } \text{CH}_4} = 6.0 \times 10^{24} \text{ atoms}$$

B. How many formula units of  $\text{CaCl}_2$  are present in 0.543 mol  $\text{CaCl}_2$ ?

$$\frac{0.543 \text{ mol } \text{CaCl}_2}{1 \text{ mol } \text{CaCl}_2} \times \frac{6.022 \times 10^{23} \text{ f.u.}}{1 \text{ mol } \text{CaCl}_2} = 3.27 \times 10^{23} \text{ f.u. } \text{CaCl}_2$$

C.  $9.81 \times 10^{21}$  molecules  $\text{O}_2$  is how many moles?

$$\frac{9.81 \times 10^{21} \text{ molecules } \text{O}_2}{6.022 \times 10^{23} \text{ molecules}} \times \frac{1 \text{ mol } \text{O}_2}{6.022 \times 10^{23} \text{ molecules}} = 0.0163 \text{ mol } \text{O}_2$$

D. How many atoms are present in 0.371 mol  $\text{P}_2\text{O}_5$ ?

$$\frac{0.371 \text{ mol } \text{P}_2\text{O}_5}{1 \text{ mol } \text{P}_2\text{O}_5} \times \frac{6.022 \times 10^{23} \text{ molecules } \text{P}_2\text{O}_5}{1 \text{ mol } \text{P}_2\text{O}_5} \times \frac{7 \text{ atoms}}{1 \text{ molecule } \text{P}_2\text{O}_5} = 1.56 \times 10^{24} \text{ atoms}$$

# Chemistry 110 Practice Problems

**JS Bruner**



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**Register of the University of California** University of California, Berkeley, 1928 **General Catalogue** Massachusetts Institute of Technology, 1925 **Annual Catalogue** Massachusetts Institute of Technology, 1916 *MCAT, Medical College Admission Test* Lawrence Solomon, Morris Bramson, 1988 Including more material and better coverage than any of the competition, this completely revised and enlarged guide incorporates four full length sample tests complete with coverage of the new essay question. **Ohio State University Bulletin**, 1921 **The Journal of the Chemical, Metallurgical & Mining Society of South Africa**, 1913 *Register* University of California, Berkeley, 1928 Catalogue University of Maryland, College Park, 1927 **American Druggists' Circular and Chemical Gazette**, 1878 **The Chemical News and Journal of Industrial Science**, 1913 Summer Session University of California (1868-1952), 1928 **The Druggists' Circular and Chemical Gazette**, 1878 Includes Red book price list section title varies slightly issued semiannually 1897-1906 *General Catalog* Iowa State University, 1917

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