

ATOMIC STRUCTURE

RUTHERFORD'S MODEL

- Discovered that an atom is mostly made up of empty space
- Center of the atom is a small dense positively charged nucleus
- This nucleus holds the negatively charged particles in the atom

BOHR'S MODEL

- There are 2 main pieces of an atom:
- Nucleus
- Electrons

THE OLD MODEL OF THE ATOM



WAVE-MECHANICAL MODEL

- Electrons are moving so fast that they look like a cloud. Orbitals are regions of most likely electron location



WHAT IS TRUE TODAY...

- The nucleus is composed of 2 things:
 - PROTONS - positively charged particles
 - NEUTRONS - neutral charged particles
- Electrons are in a cloud - they are in orbitals
 - Electrons DO move around the nucleus in different patterns



| Subatomic Particle | Location in the Atom | Charge | Weight |
|--------------------|--|---------------------|------------------------|
| proton | center (in the nucleus) | positive | 1 amu |
| neutron | center (in the nucleus) | neutral (no charge) | 1 amu |
| electron | around the nucleus in clouds or orbitals | negative | Almost equal to 1/1836 |

IONS

- If an atom gains or loses an electron, it is called an ion
- This gives the atom a charge:
 - Lose an electron, the atom becomes positive
 - Gain an electron, the atom becomes negative

ISOTOPES

- An isotope is when an atom is physically the same, but different
- An isotope is still the same element, it is just a little different from every other atom of the same element
 - slightly different mass

Chemistry Atomic Structure Study Guide

Michael Hoelscher



Chemistry Atomic Structure Study Guide:

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This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder s tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl Paul B. Kelter, 1993 Resources in education , 1982-02 Chemistry Basics (Speedy Study Guide) Speedy Publishing, 2014-12-28 A study guide of chemistry basics would give students bite sized chunks of information on a complex subject making the task of learning it easier and more enjoyable Study guides present detailed information on a topic in a compact and simplistic way which helps the student grasp more of what they are being taught Also the visual aids that usually are found in study guides can be very effective in keeping the attention of the child as well as embedding the material in their memory Study Guide/Selected Solutions Manual Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book **Barron's how to Prepare for College Entrance Examinations** Samuel C. Brownstein, Mitchel Weiner, 1974 A guide to preparing for college entrance examinations with emphasis on study programs for the verbal mathematics and standard written English parts of the SAT Includes practice tests *Student Study Guide to Accompany Essentials Anatomy and Physiology* Jackie L Butler, Shier, Ricki Lewis, Dr, Nancy Ann Sickles Corbett, 1999-06 **Student Study Guide, Human Anatomy and Physiology, Fifth Edition** Nancy A. Sickles Corbett, 1990 **A Guide to the Literature of Chemistry** Evan Jay Crane, Austin McDowell Patterson, 1927 **Chemical Abstracts** , 1926
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Table of Contents Chemistry Atomic Structure Study Guide

1. Understanding the eBook Chemistry Atomic Structure Study Guide
 - The Rise of Digital Reading Chemistry Atomic Structure Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Atomic Structure Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Atomic Structure Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Atomic Structure Study Guide
 - Personalized Recommendations
 - Chemistry Atomic Structure Study Guide User Reviews and Ratings
 - Chemistry Atomic Structure Study Guide and Bestseller Lists
5. Accessing Chemistry Atomic Structure Study Guide Free and Paid eBooks
 - Chemistry Atomic Structure Study Guide Public Domain eBooks
 - Chemistry Atomic Structure Study Guide eBook Subscription Services
 - Chemistry Atomic Structure Study Guide Budget-Friendly Options
6. Navigating Chemistry Atomic Structure Study Guide eBook Formats

- ePub, PDF, MOBI, and More
- Chemistry Atomic Structure Study Guide Compatibility with Devices
- Chemistry Atomic Structure Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Atomic Structure Study Guide
 - Highlighting and Note-Taking Chemistry Atomic Structure Study Guide
 - Interactive Elements Chemistry Atomic Structure Study Guide
- 8. Staying Engaged with Chemistry Atomic Structure Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Atomic Structure Study Guide
- 9. Balancing eBooks and Physical Books Chemistry Atomic Structure Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Atomic Structure Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Atomic Structure Study Guide
 - Setting Reading Goals Chemistry Atomic Structure Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Atomic Structure Study Guide
 - Fact-Checking eBook Content of Chemistry Atomic Structure Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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