

UNIT 4: CHEMICAL REACTIONS STUDY GUIDE

Multiple choice:

C 1. Chemical reactions _____.
a. simply change the form or state of the reactants.
b. create new atoms.
c. always produce new substances.
d. create more matter than what they started with.

B 2. Which of the following is the correct skeleton equation for the combination of solid zinc with liquid bromine to produce solid zinc bromide?
a. $Zn_{(s)} + Br_{(l)} \rightarrow ZnBr_{(s)}$
b. $Zn_{(s)} + Br_{(l)} \rightarrow ZnBr_{(s)}$
c. $Zn_{(s)} + Br_{(aq)} \rightarrow ZnBr_{(s)}$
d. $ZnBr_{(s)} \rightarrow Zn_{(s)} + Br_{(l)}$

A 3. A substance that is needed to speed up a reaction but is not a product or a reactant is known as a _____.
a. catalyst b. coefficient c. spectator ion d. steroid

D 4. In a combustion reaction, one of the reactants is always _____.
a. a liquid b. water c. a metal d. oxygen

B 5. Which of the following statements is true about the decomposition of a simple binary compound?
a. The products cannot be predicted.
b. The reactant is a single substance.
c. The reactant must always be a molecular substance.
d. The product is a single substance.

C 6. The reaction $2P_2O_5 \rightarrow 4P + 5O_2$ is an example of which type of reaction?
a. double replacement c. decomposition e. synthesis
b. single replacement d. combustion

Chemistry Section Review Study Guide Answer Key

George Tyler Miller

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