

Use your solubility curve chart to answer the following questions.

Part 1: Graph interpretation: (no work needed)

What are the customary units of solubility on solubility curves? **Grams of Solute per 100 g of water**

1. According to the graph, the solubility of any substance changes as **temperature** changes.
2. Most substances on this graph show increased solubility as temperature increases. What are the exceptions? **$\text{Ca}_2(\text{SO}_4)_3$ and NH_3**
3. The solubilities of substances whose curves show greater (steeper) slopes are **(more/less)** affected by temperature changes than those that have more gradual slopes.
4. Which salt has solubility values that are least affected by changes in temperature? **NaCl**
5. What is the solubility of ammonium chloride at 60°C? **50 g/100 g H_2O**
6. At what temperature do potassium chlorate and potassium chloride have the same solubility in water? **95 °C**
7. Which compound is least soluble in water at 12°C? **KClO_3**
8. A saturated solution of which compound contains 130 grams of solute per 100 grams of water at 70°C? **KNO_3**
9. Are the following solutions unsaturated, saturated, or supersaturated?
 - a. 80 g of sodium nitrate in 100 g of water at 30°C. **unsaturated**
 - b. 80 g of potassium chlorate in 100 g of water at 50°C. **supersaturated**
10. Which saturated solution of a chloride has the greatest percentage by mass of solute at 60°C? **don't worry about this question for the test**

Problems: Show all work on a separate sheet.

Chemistry Solution Concentration Study Guide Answer Key

John A. Olmsted, Gregory M. Williams

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