

# Chemistry Spring Final Study Guide

**Unit 10, Acids & Bases**  
Fig. 10.10: Rank the acids and compare the properties of acids at 25°C.

Acid	Base
Hydrochloric acid	Sodium hydroxide
Sulfuric acid	Ammonia
Nitric acid	Calcium hydroxide
Acetic acid	Sodium acetate
Carbonic acid	Ammonium hydroxide
Phosphoric acid	Sodium phosphate
Hydrofluoric acid	Ammonium fluoride
Hydrocyanic acid	Sodium cyanide
Perchloric acid	Ammonium perchlorate
Chloric acid	Ammonium chlorate
Bromic acid	Ammonium bromate
Iodic acid	Ammonium iodate
Chlorous acid	Ammonium chlorite
Bromous acid	Ammonium bromite
Ious acid	Ammonium iodite
Chlorous acid	Ammonium chlorite
Bromous acid	Ammonium bromite
Ious acid	Ammonium iodite

**Unit 11, Solubility**  
Fig. 11.1: Rank the solubilities of the following compounds at 25°C.

Compound	Solubility
AgCl	Insoluble
PbCl <sub>2</sub>	Slightly soluble
CaCl <sub>2</sub>	Very soluble
NaCl	Very soluble
KCl	Very soluble
LiCl	Very soluble
NaNO <sub>3</sub>	Very soluble
KNO <sub>3</sub>	Very soluble
LiNO <sub>3</sub>	Very soluble
Na <sub>2</sub> CO <sub>3</sub>	Slightly soluble
K <sub>2</sub> CO <sub>3</sub>	Slightly soluble
Li <sub>2</sub> CO <sub>3</sub>	Insoluble
Na <sub>2</sub> SO <sub>4</sub>	Slightly soluble
K <sub>2</sub> SO <sub>4</sub>	Slightly soluble
Li <sub>2</sub> SO <sub>4</sub>	Insoluble
Na <sub>2</sub> SiO <sub>3</sub>	Insoluble
K <sub>2</sub> SiO <sub>3</sub>	Insoluble
Li <sub>2</sub> SiO <sub>3</sub>	Insoluble
Na <sub>2</sub> PO <sub>4</sub>	Slightly soluble
K <sub>2</sub> PO <sub>4</sub>	Slightly soluble
Li <sub>2</sub> PO <sub>4</sub>	Insoluble
Na <sub>2</sub> VO <sub>4</sub>	Slightly soluble
K <sub>2</sub> VO <sub>4</sub>	Slightly soluble
Li <sub>2</sub> VO <sub>4</sub>	Insoluble

**Unit 12, Thermodynamics**  
Fig. 12.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 13, Equilibrium**  
Fig. 13.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 14, Kinetics**  
Fig. 14.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 15, Organic Chemistry**  
Fig. 15.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 16, Biochemistry**  
Fig. 16.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 17, Environmental Chemistry**  
Fig. 17.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 18, Analytical Chemistry**  
Fig. 18.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 19, Physical Chemistry**  
Fig. 19.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 20, Miscellaneous**  
Fig. 20.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 21, Advanced Topics**  
Fig. 21.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 22, Final Review**  
Fig. 22.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 23, Appendix**  
Fig. 23.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 24, Final Exam**  
Fig. 24.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 25, Final Review**  
Fig. 25.1: Rank the following compounds at 25°C.

Compound	Rank
CH <sub>4</sub>	1
C <sub>2</sub> H <sub>6</sub>	2
C <sub>3</sub> H <sub>8</sub>	3
C <sub>4</sub> H <sub>10</sub>	4
C <sub>5</sub> H <sub>12</sub>	5
C <sub>6</sub> H <sub>14</sub>	6
C <sub>7</sub> H <sub>16</sub>	7
C <sub>8</sub> H <sub>18</sub>	8
C <sub>9</sub> H <sub>20</sub>	9
C <sub>10</sub> H <sub>22</sub>	10
C <sub>11</sub> H <sub>24</sub>	11
C <sub>12</sub> H <sub>26</sub>	12
C <sub>13</sub> H <sub>28</sub>	13
C <sub>14</sub> H <sub>30</sub>	14
C <sub>15</sub> H <sub>32</sub>	15
C <sub>16</sub> H <sub>34</sub>	16
C <sub>17</sub> H <sub>36</sub>	17
C <sub>18</sub> H <sub>38</sub>	18
C <sub>19</sub> H <sub>40</sub>	19
C <sub>20</sub> H <sub>42</sub>	20

**Unit 26,**

# Chemistry Spring Final Review Study Guide

**Teachers of English to Speakers of  
Other Languages**



## Chemistry Spring Final Review Study Guide:

*Resources in Education* ,1984      **Cornell University Courses of Study** Cornell University,1992      **New Serial Titles**  
,1999 A union list of serials commencing publication after Dec 31 1949      **Research in Education** ,1972      *Bulletin*  
American Society for Testing Materials,1929      **Journal of Education** ,1893      *The Journal of Education* ,1893  
**Energy Research Abstracts** ,1991-10      Columbia Alumni News ,1926      PSI: Personalized System of Instruction  
John Gilmour Sherman,1974      *Medical Books and Serials in Print* ,1984      **Peterson's Annual Guides to Graduate**  
**Study** ,1983      **The Educational Times, and Journal of the College of Preceptors** ,1895      Peterson's Guide to  
Graduate Programs in the Humanities, Arts, and Social Science Peterson's Guides, Inc,1992      *The Nation* ,1919      *The*  
*Athenaeum* ,1885      *Educational Screen and Audiovisual Guide* ,1957      **Michigan School Moderator** ,1888      TESOL  
Newsletter Teachers of English to Speakers of Other Languages,1981      *Journal of the First-year Experience & Students in*  
*Transition* ,2016

This book delves into Chemistry Spring Final Review Study Guide. Chemistry Spring Final Review Study Guide is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chemistry Spring Final Review Study Guide, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Chemistry Spring Final Review Study Guide
- Chapter 2: Essential Elements of Chemistry Spring Final Review Study Guide
- Chapter 3: Chemistry Spring Final Review Study Guide in Everyday Life
- Chapter 4: Chemistry Spring Final Review Study Guide in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Chemistry Spring Final Review Study Guide. The first chapter will explore what Chemistry Spring Final Review Study Guide is, why Chemistry Spring Final Review Study Guide is vital, and how to effectively learn about Chemistry Spring Final Review Study Guide.
3. In chapter 2, this book will delve into the foundational concepts of Chemistry Spring Final Review Study Guide. This chapter will elucidate the essential principles that must be understood to grasp Chemistry Spring Final Review Study Guide in its entirety.
4. In chapter 3, the author will examine the practical applications of Chemistry Spring Final Review Study Guide in daily life. This chapter will showcase real-world examples of how Chemistry Spring Final Review Study Guide can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Chemistry Spring Final Review Study Guide in specific contexts. This chapter will explore how Chemistry Spring Final Review Study Guide is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Chemistry Spring Final Review Study Guide. This chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Chemistry Spring Final Review Study Guide.

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### Chemistry Spring Final Review Study Guide Introduction

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