

Basic Chemistry Study Guide 2

1. Hydrogen bonds

1. Form between two polar molecules **covalent bonds**
2. A hydrogen's δ^+ region is attracted to a δ^- region in another molecule.
3. Form between water molecules (Where are the δ^+ and δ^- regions in a water molecule? **hydrogen bonds, at the top of the symbol**)
4. Each hydrogen bond is very weak (can be easily formed and broken).
5. There are many hydrogen bonds in water. Together, these bonds give water its unique properties that are essential for life
6. One of these properties is the ability to be a solvent for other polar molecules.
7. Hydrogen bonds are depicted in figures as dotted or dashed lines between two atoms.

2. Solutions, solvents, and solutes

1. Water is the most abundant molecule in living organisms, and 60% of human body weight is water.
2. Water is located inside and outside of our cells.
3. Body fluids are solutions
4. Define the terms **solution, solvent, and solute**. **Solvent: a substance capable of dissolving other polar molecules and ionic bonds, solutes: molecules dissolved in the solvent, solution: homogeneous mixture**

3. How substances dissolve in water

1. When a solute dissolves, it becomes surrounded by water molecules.
2. For a solute to dissolve, it must attract water molecules.
3. What kinds of substances attract water molecules? **Sugars, nucleic acids, and amino acids**
4. When a solute dissolves it forms hydrogen bonds with water molecules.
5. Define the terms **ion, cation, and anion**. **ion: atom with a charge, cation: positively charged ion, anions: negatively charged ion**
6. Define the terms **hydrophilic** and **hydrophobic** **Hydrophilic: attracted to water, hydrophobic: afraid of water**
7. What types of molecules are hydrophilic and what types are hydrophobic? **Nonpolar hydrophobic, polar hydrophilic**

Chemistry Study Guide Completion

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Zafar, Moaz Siddiqui, Rachna Rani, Reetika Gulati, Sonal Chauhan, Maukta Gigras, 2021-11-25

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Physics and Measurement Kinematics Laws of Motion Work Energy and Power Rotational Motion Gravitation Properties of Solids Mechanical Properties of Fluids Thermal Properties of Matter Thermodynamics Kinetic Theory of Gases Simple Harmonic Motion Wave Motion Electrostatics Capacitance Current Electricity Magnetic Effects of Current Magnetism EM Induction and AC electromagnetic Waves Ray Optics Wave Optics Dual Nature of Matter and Radiation Atoms Nuclear Physics and Radioactivity Electronic Devices Communication Systems

Chemistry

Matter and Laws of Chemical Combinations Chemical Equations and Stoichiometry States of Matter Gaseous and Liquid States States of Matter Solid State Atomic Structure Radioactivity and Nuclear chemistry Chemical Bonding and Molecular Structure Chemical Thermodynamics Solutions Chemical Equilibrium Ionic Equilibrium Redox Reactions Electrochemistry Chemical Kinetics Adsorption Colloidal State Periodic Classification and Periodic Properties Principles and Process of Metallurgy

Hydrogen s p d f Block Elements Coordination Compounds Environmental Chemistry Purification of Organic Compounds Some Basic Principles of Organic Chemistry Hydrocarbons Organic Compounds Containing Halogens Alcohols Phenols and Ether Aldehyde Ketones and Carboxylic Acid Organic Compounds Containing Nitrogen Polymers Biomolecules

Chemistry in Everyday Life

Biology

The Living World Biological Classification Plant Kingdom Animal Kingdom Morphology of Flowering Plants Anatomy of Flowering Plants Structural Organization in Animals Cell Biomolecules Cell Cycle and Cell Division Transport in Plants Mineral Nutrition Photosynthesis in Higher Plants Cellular Respiration Plant Growth and Development Digestion and Absorption Breathing and Exchange of Gases Body Fluids and Circulation Excretion in Animals Locomotion and Movement Neural

Control and Coordination Endocrine System Reproduction in Organisms Social Reproduction in Flowering Plants Human Reproduction Reproductive Health Heredity and Variation Molecular Basis of Inheritance Evolution Human Health and Diseases Strategies for Enhancement in Food Production Microbes in Human Welfare Biotechnology Biotechnology and Its Application Organisms and Population Ecosystem Biodiversity and Its Conservation Environmental Issues **Revise As/A2 Biology** Senior Lecturer in African History John Parker, HarperCollins UK, 2008-10 Level A Level Subject Biology Revise for AS A2 Biology with confidence Providing complete study support throughout the two A Level years this Biology study guide matches the curriculum content and provides in depth course coverage plus invaluable advice on how to get the best results in the exams Providing plenty of exam practice and frequent progress checks and questions to consolidate learning this AS A2 Biology study guide contains invaluable advice and preparation for the exam Included in this book examiner s tips that reveal how to achieve higher marks information presented in a clear and easy to use format exam board labels that allow students to identify content relevant to their course highlighted key points and examiner s hints to offer guidance progress check questions to test recall and understanding sample questions and model answers that reveal what examiners are looking for exam style questions and answers that provide crucial exam practice eal what examiners are looking for exam style questions and answers that provide crucial exam practice eal what examiners are looking for exam style questions and answers that provide crucial exam practice *Complete Physics for NEET(UG) Medium-English* GURCHARANAM ACADEMY PRIVATE LIMITED, Complete Physics Class 11th 12th for NEET UG Medium English **Study Guide with Selected Solutions** Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems *A Complete Guide to Matriculation at the University of London* William Dodds, 1882 *Complete Graded Arithmetic* George Edward Atwood, 1903 *Manual of Blowpipe Analysis, Qualitative and Quantitative, with a Complete System of Determinative Mineralogy* Henry Bedinger Cornwall, 1902 A New and Complete Set of Traverse Tables John Theophilus Boileau, 1904 Complete Chemistry Study Guide + Flashcards and MCAT Review Konstantinos Papadopoulos, 2013-05-14 This book is part of The Dean s Guide series including study guides on biology chemistry physics and organic chemistry A portion of the proceeds from each book will go to charity sponsoring a less fortunate child The author of the Dean s Guide to Chemistry is a student himself who has chosen to write a simplified version of chemistry that can be used for chemistry at any level high school AP chemistry college and even the MCAT The author has published scientific publications on malaria and is currently pursuing his M S at the University of Glasgow in Infection Immunobiology researching under the director of the institute on rheumatoid arthritis The author uses simple real life examples and illustrations to allow students at any level to understand chemistry The study guide is based on a simplified

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