

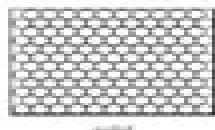
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Chapter 1

Exercise 1.1

1 A solid has a fixed mass and volume / shape.
A liquid has a fixed mass but its shape changes to that of the container in which it is placed.
A gas has no fixed shape or volume. A gas completely fills the container that it is in.

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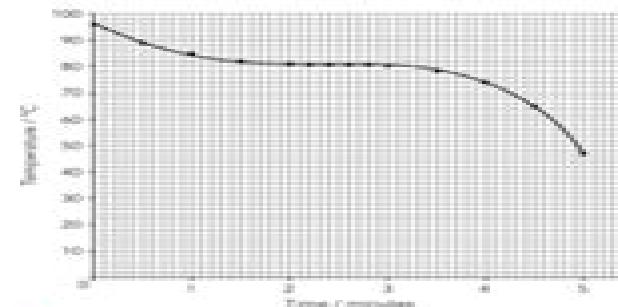
3 a B
b C
c A
d D

4 A freezing (solidification)
B melting
C condensation
D evaporation or boiling

5 a radon
b radon and nitrogen
c nitrogen
d cobalt
e The sample of ethanoic acid is impure.
The presence of impurities raises the
boiling point of a substance.

Exercise 1.2

10



- 7 The student would need to use an oil bath (in place of the water bath) so that the higher temperature could be reached.
- 8 The substance is freezing (solidifying) / turning from liquid to solid.
- 9 The temperature stays constant because energy is being released as the substance solidifies / the molecules are giving out heat as they stop moving from place to place and become organised in a structured lattice arrangement / in the solid the molecules can only vibrate about fixed points / the heat released by the formation of new interactions (forces) between the particles keeps the temperature constant until all the substance is solid.

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Xiang Xie

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