

# Thermochemistry Study Guide

**Qualification:** Write the Scientific Notation & use it to answer the questions below

Scientific Notation

1. 100 grams of a substance at 100°C, calculate

2. 100 g of water at 100°C, calculate

**Qualification:** Calculate

100 g of water

100°C



3. How much energy

4. Calculate the energy

**Qualification:** Answer the

1. 100 g of water at

2. 100 g of water at

The specific heat

Name \_\_\_\_\_ Period \_\_\_\_\_ Date \_\_\_\_\_

## Thermochemistry Study Guide

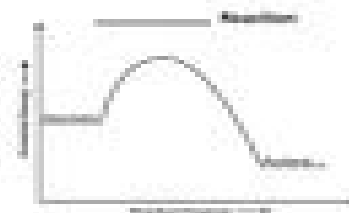
Fill in the blanks with the correct word from the following word bank to complete the statements.  
(Word Bank: heat, chemical equation, light, thermodynamics, energy, temperature, transformation, chemical)

1. According to the law of \_\_\_\_\_, energy cannot be created or destroyed, only \_\_\_\_\_ from one form to another.
2. \_\_\_\_\_ is the ability to do work. \_\_\_\_\_ energy is stored in chemical bonds. \_\_\_\_\_ energy is due to the motion of particles and is felt as heat. \_\_\_\_\_ energy is one we can use and is created by electromagnetic waves.
3. \_\_\_\_\_ is the measurement of heat energy which can be measured using thermometers while \_\_\_\_\_ is the energy transferred due to temperature difference.
4. There is thermal \_\_\_\_\_ when both the system and the surroundings have the same temperature.

Identify which Potential Energy Diagram is Endothermic and which is Exothermic and fill in the blanks to compare their properties.



Heat is \_\_\_\_\_  
\_\_\_\_\_ Chemical Bonds  
Q is \_\_\_\_\_  
Example: \_\_\_\_\_



Heat is \_\_\_\_\_  
\_\_\_\_\_ Chemical Bonds  
Q is \_\_\_\_\_  
Example: \_\_\_\_\_

# Chemistry Study Guide For Equilibrium And Thermochemistry

**K Payea**



## **Chemistry Study Guide For Equilibrium And Thermochemistry:**

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**Chemistry, Study Guide** Bernice G. Segal, Peter S. Shenkin, 1989-02-14 This Second Edition of the first year chemistry text known for its clarity of exposition and its large number of illustrative worked problems contains a more rigorous treatment of electrochemistry chemical equilibrium and thermochemistry. Worked examples now number over 300 and exercises over 1460

**Journal of the American Chemical Society** American Chemical Society, 1927 Proceedings of the Society are included in v 1 59 1879 1937

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