

Thermochemistry Study Guide

Instructions: Write the answers next to the questions & use it to answer the questions below.

Answers:
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Name: _____ Date: _____

Thermochemistry Study Guide

Use the answers with the correct word from the following word bank to complete the questions. (Note: heat, endotherm, exotherm, heat capacity, energy, temperature, enthalpy, thermal energy)

- According to the law of _____, energy cannot be created or destroyed.
- _____ is the ability to do work. _____ energy is stored in chemical bonds.
- _____ energy is stored in the motion of particles and is transferred by _____.
- _____ is the measurement of heat energy when it has measured using a thermometer.
- Heat is defined _____, when both the system and the surroundings have the same temperature.

Identify each Potential Energy Diagram as Endothermic and which is Exothermic and fill in the boxes to complete their properties.

Reaction

Reaction: _____

Chemical Bonds: _____

Q: _____

Exotherm: _____

Reaction

Reaction: _____

Chemical Bonds: _____

Q: _____

Exotherm: _____

Chemistry Study Guide For Equilibrium And Thermochemistry

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Chemistry, Study Guide Bernice G. Segal, Peter S. Shenkin, 1989-02-14

This Second Edition of the first year chemistry text known for its clarity of exposition and its large number of illustrative worked problems contains a more rigorous treatment of electrochemistry, chemical equilibrium and thermochemistry. Worked examples now number over 300 and exercises over 1460.

Journal of the American Chemical Society American Chemical Society, 1927

Proceedings of the Society are included in v 1-59 (1879-1937).

The Elements of Physical Chemistry Harry Clary, Jones, 1903

A Laboratory Manual in General Chemistry William Foster, 1923

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